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<b>(21) International Application Number:</b> PCT/US96/08994 <b>(22) International Filing Date:</b> 6 June 1996 (06.06.96)  <b>(30) Priority Data:</b> <table border="0" style="width: 100%;"><tr><td style="width: 40%;">08/477,107</td><td style="width: 40%;">7 June 1995 (07.06.95)</td><td style="width: 20%;">US</td></tr><tr><td>08/488,297</td><td>7 June 1995 (07.06.95)</td><td>US</td></tr><tr><td>08/488,300</td><td>7 June 1995 (07.06.95)</td><td>US</td></tr></table> <b>(71) Applicant:</b> ARIZONA BOARD OF REGENTS [US/US]; Arizona State University, Tempe, AZ 85287 (US).  <b>(72) Inventor:</b> NELSON, Randall, W.; Apartment 1051, 11221 South 51st Street, Phoenix, AZ 85044 (US).  <b>(74) Agent:</b> MacBLAIN, Thomas, D.; Brumbaugh, Graves, Dono- hue & Raymond, 30 Rockefeller Plaza, New York, NY 10112 (US).		08/477,107	7 June 1995 (07.06.95)	US	08/488,297	7 June 1995 (07.06.95)	US	08/488,300	7 June 1995 (07.06.95)	US	<b>(81) Designated States:</b> AU, CA, JP, European patent (AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE).  <b>Published</b> <i>With international search report.</i> <i>With amended claims.</i>
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<b>(54) Title:</b> A SAMPLE PRESENTATION APPARATUS FOR MASS SPECTROMETRY  <b>(57) Abstract</b>  A sample presentation device, with a surface-bound complex including at least one molecule which chemically modifies a biomolecule, is prepared and exposed to a biomolecule. The molecular weight of the chemically modified biomolecule is then determined by mass spectrometry.											

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### Description

#### A Sample Presentation Apparatus For Mass Spectrometry

#### Background of the Invention

The present invention relates generally to derivatized mass spectrometry sample presentation apparatuses, and more specifically, to mass spectrometry sample presentation apparatuses  
5 derivatized with complexes including at least one molecule which modifies a biomolecule.

Conventionally, mass spectrometry is a technique used to characterize analytes by determining their molecular weight. Ordinarily, mass spectrometry  
10 involves the steps of: coating a sample presentation apparatus with an analyte, introducing the sample presentation apparatus into the mass spectrometer, volatilizing and ionizing the analyte, accelerating the ionized analyte toward a detector by exposing the ions  
15 to an electric and/or a magnetic field, and analyzing the data to determine the mass to charge ratio of specific analyte ions.

If an analyte remains intact throughout this process, data will be obtained which corresponds to a  
20 molecular weight for the entire intact analyte ion. Typically however, it is beneficial to additionally obtain data corresponding to the molecular weight of various fragments of the analyte. It is also beneficial to obtain data which only corresponds to the  
25 pure analyte, even when impurities are present. Therefore, it is advantageous to be able to modify analytes by purifying and/or cleaving them, prior to determining their molecule weight.

In conventional mass spectrometry, the analyte is  
30 either modified before it is coated on the sample

presentation apparatus, or during the volatilization and ionization steps, which occur inside the mass spectrometer. It is known that biomolecules (e.g. polypeptides, deoxyribonucleic acid (DNA), ribonucleic acid (RNA), or carbohydrates) can be selectively digested at specific locations by exposing them to immobilized complexes. However, it is disadvantageous to modify an analyte before it is coated on the sample presentation apparatus, as this extra step slows the overall process, involves loss of the analyte, and can possibly introduce contaminants. Moreover, if the fragments are generated in solution in a reaction between the analyte and a reagent, the kinetics of the reaction may be rather slow, thus adding further delay. It is also disadvantageous to generate fragments during the volatilization and ionization steps as such methods typically provide little control over the analyte cleavage site and may lead to excessive degradation of the analyte.

It is further known to use the Matrix-Assisted Laser Desorption/Ionization ("MALDI") technique to volatilize and ionize biomolecules in a mass spectrometer. This technique involves surrounding a biomolecule in a special matrix material. A LASER beam, tuned to a frequency where the matrix absorbs, is targeted on the matrix material. The LASER transfers sufficient energy to volatilize a small portion of the matrix material. A small number of analyte molecules are thus carried along with the matrix material into the vapor phase in the mass spectrometer.

Prior to the development of MALDI, analysis of biomolecules by mass spectrometry was quite difficult, if not impossible, since no techniques were available which were gentle enough to volatilize intact biomolecules without any degradation or fragmentation. While the MALDI technique provides an advantageous

technique for volatilizing biomolecules, this technique does not provide for generation of analyte fragments or purification of an analyte prior to introducing the analyte onto a sample presentation apparatus. In

5 particular, the need for extra fragmentation or purification steps slow the overall process, wastes valuable analyte, and can introduce contaminants.

It is known from PCT Publication No. 94/28418 to use a sample presentation apparatus for laser  
10 desorption and ionization mass spectrometry, wherein the sample presentation apparatus is derivatized with surface associated molecules for promoting desorption and ionization of biomolecules. In such surface-bound biomolecule methods, the surface associated molecules  
15 bind biomolecules, and later promote desorption of the biomolecules when they are exposed to LASER radiation. Biomolecule fragments are generated by exposing the surface-bound biomolecules to reagents in solution, and later washing away the reagent. This procedure is  
20 problematic because it can introduce contaminants, which reduces the quality of the collected data. Specifically, if any reagent is left on the surface after the washing procedure, it will be the first material to be volatilized by the LASER beam, and may  
25 very well swamp the signals from the material of interest, thereby adversely affecting the data.

Moreover, such a process is difficult because it relies on new and unproven systems for volatilizing biomolecules. In contrast, MALDI matrix systems are  
30 well studied, and known to effectively and gently volatilize biomolecules. MALDI matrices cannot be used with the surface-associated molecule method because the matrix covers the surface-bound biomolecules, and thus would completely prevent the acquisition of useful  
35 data.

Furthermore, it is also disadvantageous to directly bind biomolecules to a sample presentation surface, because this would eliminate the possibility of transferring a modified biomolecule to a second region on the presentation surface or a second presentation surface, where it could be exposed to other modifying reagents. Therefore, any chemical reactions according to the surface-bound analyte method, must be performed sequentially, rather than in parallel. In such sequential methods, the bound analyte must be reacted with one reagent, introduced into the mass spectrometer, removed, and then subsequently reacted with a second reagent. Each of these steps is disadvantageous as it presents an opportunity for contaminating the sample and introduces further delay.

For the foregoing reasons, there is a need for a mass spectrometry sample presentation apparatus which provides a rapid, efficient method for analyzing biomolecules using proven, advantageous MALDI matrices.

#### Summary of the Invention

Accordingly, it is an object of the invention to provide a sample presentation apparatus, mass spectrometry method, and mass spectrometer that overcome the disadvantages of the prior art.

Another object of the invention is to provide an improved sample presentation apparatus including a complex capable of modifying biomolecules wherein it is unnecessary to separate any reagent from the modified biomolecule to prevent contaminating data obtained from the modified biomolecule.

Another object of the invention is to provide an improved derivatized mass spectrometry sample presentation apparatus which minimizes sample loss, and

thus works effectively with extremely small amounts of biomolecules.

Yet another object of the invention is to provide a mass spectrometry sample presentation apparatus which presents a complex capable of chemically modifying a biomolecule in a high effective concentration, and thus increases the rate of biomolecule modification reactions.

Another object of the invention is to provide a mass spectrometry sample presentation apparatus which digests and/or purifies biomolecules.

A further object of the invention is to provide a derivatized mass spectrometry sample presentation apparatus which can utilize existing optimized MALDI matrix systems.

Another object of the invention is to provide a mass spectrometry method wherein biomolecules are reacted with one reagent, and then a portion of the reacted biomolecules are reacted with a different reagent.

A further object of the invention is to provide a method for making a mass spectrometry sample presentation apparatus which overcomes the disadvantages of the prior art.

Yet another object of the invention is to provide a method of analyzing a biomolecule, wherein the biomolecule is exposed to the derivatized surface of the sample presentation apparatus and then incubated in a moist atmosphere. Subsequently, a MALDI matrix material is added, and finally the sample presentation apparatus is introduced into the mass spectrometer.

Another object of the invention is to provide a method of analyzing a polypeptide, wherein the molecular weight data is used to determine a limited amino acid sequence of the peptide. The limited amino acid sequence obtained by this method can then be used

to generate corresponding nucleic acid sequences. These nucleic acid sequences can then be used to search a computer database containing known nucleic acid sequences. A match may identify a known nucleic acid sequence which codes for the polypeptide.

A further object of the invention is to provide an improved mass spectrometer which incorporates an improved mass spectrometry sample presentation apparatus.

These and other objects of the present invention are obtained by providing a mass spectrometry sample presentation apparatus comprising a mass spectrometry sample presentation surface, wherein a complex is bound to the surface which includes at least one molecule which chemically modifies a biomolecule.

A method of making a sample presentation apparatus is further provided wherein a complex, which includes at least one molecule which chemically modifies a biomolecule, is bound to the sample presentation apparatus surface.

The objects of the present invention are further accomplished through a method of analyzing a biomolecule comprising the steps of:

- (a) providing a mass spectrometry sample presentation surface;
- (b) binding at least one complex to the surface wherein the complex includes at least one molecule which chemically modifies a biomolecule;
- (c) contacting the biomolecule with the surface, thereby chemically modifying the biomolecule; and
- (d) determining the molecular weight of the chemically modified biomolecule in a mass spectrometer.



The objects of the present invention are further obtained by a mass spectrometer. The mass spectrometer has a sample presentation surface with a surface-bound complex including at least one molecule which  
5 chemically modifies a biomolecule. The mass spectrometer also has a vacuum interlocking device for introducing the surface into the machine, an apparatus for volatilizing and ionizing the chemically modified biomolecule, an electric field generator, a detector  
10 and electronics for determining the molecular weight to charge ratio of the modified biomolecule ions.

The sample presentation apparatus, method of making a sample presentation apparatus, method of analyzing biomolecules and mass spectrometer of the  
15 present invention provide mass spectrometry information for biomolecules which is both easier to obtain and of higher quality, than was previously possible.

#### Brief Description of the Drawings

Fig. 1 is a MALDI mass spectrum of 10 picomoles of  
20 chicken egg lysozyme digested for 10 minutes with surface-bound trypsin.

Fig. 2 is a summary illustrating the correlation between the calculated and observed masses for chicken egg lysozyme fragments.

25 Fig. 3 is a MALDI mass spectrum of 1 picomole of chicken egg lysozyme digested for 10 minutes with surface-bound trypsin.

Fig. 4 is a MALDI mass spectrum of 1 picomole of chicken egg lysozyme digested for 10 minutes with an  
30 equimolar amount of unbound trypsin.

Fig. 5 is a MALDI mass spectrum of 2  $\mu$ l of a 500  $\mu$ M solution of chicken egg lysozyme digested with an equal volume of slurried agarose immobilized trypsin.

Fig. 6 is a summary of the residues of 1 picomole of chicken egg lysozyme obtained following a 10 minute digest with free trypsin and agarose-immobilized trypsin.

5        Fig. 7 is a schematic illustration of the amino acid sequence of  $\alpha$ -cobratoxin.

Fig. 8 is a MALDI mass spectrum of undigested  $\alpha$ -cobratoxin.

10       Fig. 9 is a MALDI mass spectrum of approximately 10 picomoles of  $\alpha$ -cobratoxin digested for 30 minutes with surface-bound trypsin sample presentation apparatus.

15       Fig. 10 is a MALDI mass spectrum of approximately 10 picomoles of  $\alpha$ -cobratoxin digested for 30 minutes with a surface-bound trypsin sample presentation apparatus in the presence of a reducing agent.

Fig. 11 is an exploded view of the MALDI mass spectrum of Fig. 10 in the  $m/z$  range of 1,000-5,000 Pa.

20       Fig. 12 is a positive ion MALDI mass spectrum of undigested human parathyroid hormone 1-34 (hPTH).

Fig. 13 is a MALDI mass spectrum of hPTH digested for 10 minutes on a surface-bound  $\alpha$ -chymotrypsin mass sample presentation apparatus.

25       Fig. 14 is a MALDI mass spectrum of hPTH digested for 10 minutes on a surface-bound  $\alpha$ -chymotrypsin sample presentation apparatus followed by pH adjustment to approximately 5 and addition of carboxypeptidase P (cpp). The inset illustrates the region of interest for final residue determination.

30       Fig. 15 illustrates the partial peptide sequence of hPTH generated by analyzing the mass spectrometry data from  $\alpha$ -chymotrypsin and  $\alpha$ -chymotrypsin/carboxypeptidase P (cpp) fragments.

35       Fig. 16 is a MALDI mass spectrum of hPTH digested for 5 minutes on a surface-bound trypsin sample presentation apparatus maintained at 45°C, pH 8.1.

Fig. 17 is a protein ladder sequence generated by addition of cpp to the digested sample of Fig. 16.

Fig. 18 is a summary of the partial peptide sequence generated by analyzing the mass spectrometry data from trypsin and trypsin/cpp fragments.

Fig. 19 is an overlay of the mass spectra of trypsin and trypsin/cpp fragments illustrating the use of both the large-scale and ladder information to extend the sequence determination.

Fig. 20 is a MALDI mass spectrum of hPTH digested for 8 minutes on a surface-bound pepsin sample presentation apparatus maintained at 45°C, pH 3.4.

Fig. 21 is a MALDI mass spectrum of hPTH digested for 10 minutes on a surface-bound  $\alpha$ -chymotrypsin mass sample presentation apparatus maintained at 45°C, pH 3.4.

Fig. 22 is a MALDI mass spectrum of hPTH digested for 10 minutes on a surface-bound  $\alpha$ -chymotrypsin mass sample presentation apparatus maintained at 45°C, pH 3.4, followed by a 10 minute digestion by surface-bound cpp maintained at 35°C, pH 5.

#### Description of the Preferred Embodiments

The present invention provides a conventional mass spectrometry sample presentation apparatus which has a surface derivatized with complexes which chemically modify biomolecules. This apparatus is typically exposed to a biomolecule, and incubated to allow the complex to modify the biomolecule. After incubation, the modified biomolecule is preferably surrounded with a MALDI matrix material and then introduced into a mass spectrometer. The invention also include further embodiments such as a mass spectrometer incorporating such a sample presentation apparatus. Further embodiments provide a method of performing mass

spectrometry, and of making a sample presentation apparatus.

As used herein, the term "derivatized" is meant to refer to a mass spectrometry sample presentation apparatus having a bound complex. A derivatized sample presentation apparatus in accordance with the invention may be composed of any suitable material. The material can be a solid or liquid. Suitable solid materials include, but are not limited to insulators such as quartz, semiconductors such as doped silicon and the like, and conductors including metals such as steel, gold and the like. Various insulating or conductive polymers may also be used. The surface of the sample presentation apparatus need not be made of the same material as the rest of the apparatus. It is preferable for the surface to be clean so that a complex may adhere to the surface.

The complex attached to the presentation apparatus surface has two distinct functionalities. As used herein, the term "complex" refers to a chemical which includes at least one molecule which chemically modifies a biomolecule. The complex has a "tethering" function, wherein it is capable of attaching itself at at least one end to the surface. Moreover, it also has a "reactive" function wherein it is capable of chemically modifying the analyte. As used herein, the term "chemically modify" is meant to refer to purification or digestion of an analyte, but not binding an analyte. The complex may further comprise a tethering molecule which binds to the surface, and a reactive molecule which binds to the tethering molecule and modifies a biomolecule.

Typically, the "tethering" function is associated with one group of atoms (i.e., the tethering molecule) while the "reactive" function is associated with a different group of atoms (i.e., the reactive molecule).

The complex may comprise both a tethering molecule which binds to the sample presentation surface and a reactive molecule which binds to the tethering molecule and modifies a biomolecule. However this separation of function is not necessary. The tethering molecule is preferably selected from the class of molecules which has proven to be effective as an immobilized ligand. Many specific examples of such molecules are listed on pages T-156 to T-200 of the 1995 Pierce Corporation Catalog (Pierce Corp., P.O. Box 117, Rockford IL 61105). Suitable tethering molecules include: dithiothreitol, dimethyladipimide-2\*HCL, dimethylpimelimide\*HCL, dimethylsuberimide\*2HCL, dimethyl 3,3'-dithiobispropionimide\*2HCL, disuccinimidyl glutarate, disuccinimidyl suberate, bis(sulfosuccinimidyl) suberate, dithiobis(succinimidylpropionate), dithiobis(sulfosuccinimidylpropionate), ethylene glycobis(succinimidylsuccinate), ethylene glycobis(sulfosuccinimidylsuccinate), disuccinimidyl tartarate, disulfosuccinimidyl tartarate, bis[2-(succinimidylloxycarbonyloxy)ethyl]sulfone, bis[2-(sulfosuccinimidooxy-carbonyloxy)ethyl]sulfone, succinimidyl 4-(N-maleimido-matyl) cyclohexane-1-carboxylate, sulfo-succinimidyl 4-(N-maleimidomethyl) cyclohexane-1-carboxylate, m-Maleimidobenzoyl-N-hydroxysuccinimide ester, m-maleimidobenzoyl-N-hydroxysulfosuccinimide ester, succinimidyl 4-p-maleimido-phenyl)-butyrate, sulfosuccinimidyl 4-(p-maleimidophenyl)-butyrate, bismaleimidohexane, N-(λ-maleimidobutyryloxy)succinimide ester, N-(λ-maleimidobutyryloxy)sulfosuccinimide ester, n-succinimidyl(4-iodoacetyl)aminobenzoate, sulfosuccinimidyl(4-iodoacetyl)-aminobenzoate, 1,4-di-[3'-2'-pyridyldithio(propionamido)butane], 4-succinimidyl-oxycarbonyl-α(s-pyridyldithio)toluene,

- sulfosuccinimidyl-6-[ $\alpha$ -methyl- $\alpha$ -(2-pyridyldithio)-toluamido]hexanoate, n-succinimidyl-3(2-pyridyldithio)-propionate, succinimidyl 6-[3-(2-pyridyldithio)-propionamido]hexanoate, sulfosuccinimidyl-6-[3-(2-pyridyldithio)-propionamido]hexanoate,
- 5 sulfosuccinimidyl-6-[3-(2-pyridyldithio)-propionamido]hexanoate, 3-(2-pyridyldithio)-propionyl hydrazine, 1-ethyl-3-(3-dimethylaminopropyl)-carbodiimide hydrochloride, n,n'-
- 10 dicyclohexylcarbodiimide, 4-(p-azidosalicylamido)-butylamine, azidobenzoyl hydrazine, N-5-azido-2-nitrobenzoyloxysuccinimide, n-5-azido-2-nitrobenzoyloxysuccinimide, n-[4-(p-azidosalicylamido)butyl]-3'(2'-
- 15 pyridyldithio)propionamide, p-azidophenyl glyoxal monohydrate, 4-(p-azidosalicyl-amido)butylamine, 1-(p-azidosalicylamido)-4-(iodoacetamido)butane, bis-[ $\beta$ -4-azidosalicylamido)ethyl]disulfide, n-
- hydroxysuccinimidyl-4-azidobenzoate, n-hydroxysulfosuccinimidyl 4-azidobenzoate, n-hydroxysuccinimidyl-4-
- 20 azidosalicylic acid, n-hydroxysulfosuccinimidyl-4-azidosalicylic acid, sulfosuccinimidyl-(4-azidosalicylamido)-hexanoate, p-nitrophenyl-2-diazo-3,3,3-trifluoropropionate, 2-diazo-3,3,3-trifluoro-
- 25 propionate, n-succinimidyl-(4-azidophenyl)1,3/-dithiopropionate, sulfosuccinimidyl-(4-azidophenyldithio)propionate, sulfosuccinimidyl-2-(7-azido-4-methylcoumarin-3-acetoamide)ethyl-1,3'-
- dithiopropionate, sulfosuccinimidyl 7-azido-4-
- 30 methylcoumarin-3-acetate, sulfosuccinimidyl 2-(m-azido-o-nitrobenzamido)-ethyl-1,3'-dithiopropionate, n-succinimidyl-6-(4'-azido-2'-nitrophenyl-amino)hexanoate, sulfosuccinimidyl 6-(4'-azido-2'-
- nitrophenylamino)hexanoate, sulfosuccinimidyl 2-(p-
- 35 azidosalicylamido)ethyl-1,3'-dithiopropionate, sulfosuccinimidyl 4-(p-azidophenyl)-butyrate. Various

protocols for using such ligands are explained in "Immobilized Affinity Ligand Techniques" by Greg T. Hemanson et al. (1992), which is incorporated herein by reference.

5       The complex may chemically modify a biomolecule  
analyte in several different ways. As used herein, the  
term "biomolecule" is meant to refer to DNA, RNA,  
polypeptides, carbohydrates and combinations thereof.  
For example, the complex may digest, or purify the  
10   biomolecule. As used herein, the term "digest" is  
meant to refer to chemical or enzymatic cleavage of a  
biomolecule into fragments and the term "purify" means  
to remove undesirable materials from a solution  
containing the biomolecule. As used herein, the term  
15   "fragment" is meant to refer to a part of a biomolecule  
which has been cleaved from the rest of the biomolecule  
using a chemical or enzymatic reagent. Several  
different reactive molecules are available for  
digesting biomolecules. If the biomolecule is a  
20   polypeptide, the complex will preferably comprise a  
chemical cleavage agent or an enzymatic protease.

Enzymatic proteases are specific polypeptides  
which cleave polypeptides. Proteases may cleave  
themselves by a process known as autolysis. Several  
25   enzymatic proteases cleave polypeptides between  
specific amino acid residues. Examples of proteases  
which cleave nonspecifically include subtilisin, papain  
and thermolysin. Examples of proteases which cleave at  
least somewhat specifically include: aminopeptidase-M,  
30   carboxypeptidase-A, carboxypeptidase-P,  
carboxypeptidase-B, carboxypeptidase-Y, chymotrypsin,  
clostripain, trypsin, elastase, endoproteinase Arg-C,  
endoproteinase Glu-C, endoproteinase Lys-C, factor Xa,  
ficin, pepsin, plasmin, staphylococcus aureus V8  
35   protease, proteinkinase K and thrombin.

Chemical cleavage agents which cleave polypeptides between specific amino acid residues include cyanogen bromide, O-iodosobenzoate or O-iodosobenzoic acid, dilute hydrochloric acid, N-bromosuccinimide, sodium hydrazine, lithium aluminum hydride, hydroxylamine and 2-nitro-5-thiocyanobenzoate. Suitable chemical cleavage agents also include Sanger's Reagent (2,4-dinitrofluorobenzene), Edmund's Reagent (Phenylisothiocyanate), and immobilized derivatives of these Reagents. Moreover, coordinated transition metal complexes such as the tetradentate Co (III) complex,  $\beta$ -[Co(triethylenetetramine)-OH(H<sub>2</sub>O)], may also be used as effective cleavage agents. See Buckingham et al. 89 J.A.C.S. 1082 (1967).

When the biomolecule to be analyzed is a polypeptide, it may be advantageous to add a reducing agent to sever any disulfide bonds between the amino acids. Severing these bonds will result in further fragmentation of the peptide, thus yielding more information about peptide structure. Disulfide reduction agents may be bound onto the sample presentation surface, or used in solution. Suitable disulfide reducing agents include  $\beta$ -mercaptoethanol, cysteine, and dithiothreitol (DTT). Carboxypeptidase-P, dithioerythritol (DTE) lipomide and N-acetylhomocysteine can be used as surface-bound disulfide reduction agents.

For biomolecules such as DNA or RNA, it may be advantageous to add a restriction endonuclease or exonuclease to the sample presentation apparatus surface. These enzymes can cleave DNA and RNA between specific nucleases, creating fragments of DNA or RNA. Specific restriction endonucleases and exonucleases are well known to those skilled in the art. Examples of enzymes which cleave both DNA and RNA include, but are not limited, to snake-venom phosphodiesterase and



spleen phosphodiesterase. Enzymes which cleave only DNA include, but are not limited to, deoxy-ribonuclease I and II. Ribonuclease I (pancreas) and T. (mold) cleave RNA.

5        When analyzing a biomolecule containing carbohydrate, it may be advantageous to add fucose, O- or N-glycanase, mannanase, neuraminidase, galactosidase or glucosidase to the surface. These enzymes will generate carbohydrate fragments of specific molecular  
10       weight which can be determined in the mass spectrometer.

      The complex attached to the sample presentation surface may also purify the biomolecule. For example, the complex may comprise a reactive molecule which  
15       invisibly binds common impurities such as detergents, lipids, endotoxins or proteases.

      The present invention also provides a method for making the above-described derivatized mass spectrometry sample presentation apparatus. In this  
20       method it is essential that a complex which modifies biomolecules be bound to the surface of the sample presentation apparatus. In one embodiment, the entire complex is first synthesized or purified, and then reacted with the surface. In an alternate preferred  
25       embodiment, the complex is formed by first reacting the sample presentation surface with a tethering molecule, and then later reacting the tethered molecule with a reactive molecule.

      Another embodiment of the present invention  
30       provides a method for analyzing a biomolecule. Specifically, the method involves: (1) providing a mass spectrometry sample presentation surface, (2) binding at least one complex on the surface wherein the complex includes at least one molecule which chemically  
35       modifies a biomolecule, (3) contacting the biomolecule with the surface, thereby chemically modifying the

biomolecule (4) applying a MALDI matrix to the surface and (5) determining the molecular weight of the biomolecule in a mass spectrometer. The biomolecule may be contacted with the surface while in solution or  
5 in the vapor phase.

The contacting step is preferably performed above room temperature, most preferably up to 55°C, to speed the reaction rate. However, it should not be performed at a temperature high enough to cause degradation of  
10 the analyte or the complex. For certain complexes, it may be advantageous to perform the contacting step below room temperature, most preferably above -10°C. This contacting step is allowed to proceed for a sufficient time, typically 5-30 minutes. The actual  
15 reaction time will depend on the particular complex and biomolecule. When proteases are used to digest polypeptide biomolecule, the contacting step is preferably performed in a moist atmosphere. Such an atmosphere may be created for example, by providing a  
20 water reservoir and enclosing it along with the sample presentation surface and biomolecule in a closed system.

Next, a MALDI matrix is added to the reacted biomolecule. The MALDI matrix may be any material  
25 which solubilizes biomolecules, absorbs light energy at a frequency easily accessible by a laser, and is unreactive with respect to biomolecules. Suitable matrices include nicotinic acid, pyrozinoic acid, vanillic acid, succinic acid, caffeic acid, glycerol, urea or tris buffer (pH 7.3). Preferable matrices  
30 include  $\alpha$ -cyano-4-hydroxycinnamic acid, ferulic acid, 2,5-dihydroxybenzoic acid, sinapic (or sinapinic) acid, 3,5-dimethoxy, 4-hydroxy-trans-cinnamic acid, other cinnamic acid derivatives, gentisic acid and  
35 combinations thereof.

A further embodiment of the present invention provides a mass spectrometer which incorporates the above-described sample presentation apparatus. The mass spectrometer according to the present invention further comprises a vacuum interlocking device for introducing the sample presentation apparatus into a volatilization chamber, an apparatus for volatilizing and ionizing the analyte, an optional electric field generator which is oriented to cause biomolecule ions to travel toward a detector, and electronics for determining the molecular weight to charge ratio for a particular ion based on its trajectory.

This mass spectrometer is compatible with all of the techniques normally used for volatilizing and ionizing analytes. Such techniques include, but are not limited to, electrospray ionization, Californium<sup>252</sup> plasma desorption, field desorption, fast atom bombardment, liquid secondary ion MS, laser desorption, and thermospray ionization. Preferably, the MALDI technique is used for volatilizing and analyzing biomolecules. The mechanical structures or equipment required in these techniques are well known to those skilled in the art.

The spectrometer usually contains an electric field generator which has electrically charged metal structures between the area in the spectrometer where the ions are generated and the detector. These structures must be designed such that they attract ions, but the ions can travel past them to reach the detector. Preferably, the structures are charged metal grids with apertures that allow some of the ions to pass through and drift toward the detector. The structures must be located between the ion generation area and the detector. They must also have a potential difference which acts to accelerate ions toward the detector.

Some mass spectrometers also use magnetic fields to deflect ions as they travel towards the detector. Such instruments determine the mass of an ion based on its deflection in the magnetic field. It is, however, preferable to use a time-of-flight mass spectrometer to determine the mass of biomolecules.

The electronics for determining the molecular weight to charge ratio of an ion based on its trajectory typically includes an ion impact detector, data-acquisition electronics, and data analysis electronics, such as a computer.

Many types of detectors are compatible with the present invention. Some specific examples of detectors include the Faraday cup, electron multipliers, electro-optical ion detectors and photographic emulsions.

Specific embodiments in accordance with the present invention will now be described in detail. These examples are intended to be illustrative, and the invention is not limited to the materials, methods or apparatus set forth in these embodiments.

#### Example 1

##### Preparation of Sample Presentation Apparatus

Circles of gold (Au) foil (99.9%), available from Johnson Matthey, 2.5 mm diameter and 0.01 mm thick were punched out and placed in a polyethylene microcentrifuge tube. Care was taken to avoid any contamination of the gold foil circles. The gold foil was activated according to the method of Katz. See, Katz, E.Y., J. Electroanal. Chem., Vol. 291, p. 257 (1990). The circles were treated with a crosslinking, or tethering molecule, solution of Dithiobis [succinimidylpropionate] (DSP) available from Pierce dissolved in isopropanol to approximately 0.1M. The DSP/isopropanol solution was essentially saturated at this concentration. The circles were treated with the

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DSP/isopropanol solution for fifteen minutes, with occasional mixing using a vortex mixer. This tethering molecule solution was poured off, and the gold circles were rinsed repetitively with pure isopropanol, followed by three rinses with pure ethanol. The gold circles were vacuum dried and transferred to new microcentrifuge tubes prior to treatment with a reactant molecule or biomolecule.

#### Example 2

10

##### Preparation of Sample Presentation Apparatus with Surface-Bound Trypsin

A mass spectrometry sample presentation apparatus was prepared by binding trypsin to the surface of the apparatus as follows. Trypsin type XIII: tosyl-L-phenylalanine chloromethyl ketone (TPCK) treated, from bovine pancreas available from Sigma, was dissolved in 20 mM phosphate buffer, pH 7.8 to a concentration of 1 mg/mL for use in preparation of the derivatized sample presentation apparatus. To attach the enzyme to the crosslinker or tethering molecule on the gold foil circles prepared in Example 1, the circles were incubated with a 1 mg/mL solution of enzyme in 20 mM phosphate buffer, pH 7.8 overnight in a refrigerator with occasional agitation. Following incubation, the gold foil circles were rinsed vigorously with the phosphate buffer, and subsequently with a 0.1% aqueous solution of Triton-X100, using a vortex mixer for both rinses. The gold foil circles were then vacuum dried and stored in polyethylene microcentrifuge tubes. Physical attachment of the gold foil circle surfaces to the stainless steel sample presentation apparatuses conventionally used for routine MALDI analysis was accomplished by application of a small amount of spray adhesive to the conventional sample presentation surface or probe tip, which was then pressed onto one

of the gold foil circles. Prior to attachment, the conventional stainless steel (304) mass spectrometer sample presentation apparatuses were electropolished and thoroughly rinsed in deionized water followed by  
5 ultrasonic cleaning in methanol. Even pressure was maintained on the probe tip and gold foil while the adhesive was allowed to bond. The sample presentation apparatus with bound trypsin was then used for protein digestion as described hereinbelow. In practice,  
10 multiple tips were made and stored in polyethylene tubes at room temperature until needed.

### Example 3

#### Au/trypsin Sample Presentation Surface Contacted with Lysozyme

15 Digestions were performed with the Au/trypsin active surfaces prepared in Example 2 by the application of aliquots (either 4  $\mu$ L or 400 nL) of 250 nM lysozyme hen egg solution (dissolved in 20 mM phosphate; 10 mM dithiothreitol (DTT); pH 8.1 buffer)  
20 directly to the sample presentation apparatus surfaces. The surfaces were allowed to stand in a humid enclosure maintained at 40°C. The solution volume on the surface was monitored, and additional aliquots of 1-2  $\mu$ L of phosphate buffer were added if the volume of liquid  
25 appeared to be less than 2  $\mu$ L. After 10 minutes the apparatus was removed from the humid environment, 2  $\mu$ L of the MALDI matrix,  $\alpha$ -cyano-4-hydroxycinnamic acid (prepared as a saturated solution in a solvent mixture of 2:1, 1.3% aqueous trifluoroacetic acid:acetonitrile  
30 (ACCA)), was applied and the surface was allowed to air dry.

Comparative Example 4Protein Digestion/Sample  
Preparation with Free Trypsin

5 A 1:1 mole ratio of trypsin to lysozyme was  
prepared by addition of 2  $\mu$ L of 500 nM lysozyme with  
2  $\mu$ L of 500 nM trypsin (both prepared in a 20 mM  
phosphate; 10 mM DTT; pH 8.1 buffer). The combined  
volume was placed in a 600  $\mu$ L microcentrifuge tube and  
incubated at 40°C for 10 minutes. The reaction was  
10 halted by the addition of 2  $\mu$ L of the MALDI matrix ACCA  
and the total volume of the digest/matrix mixture was  
placed on an inert mass spectrometer sample  
presentation apparatus and allowed to air dry.

Comparative Example 515 Agarose Immobilized Trypsin

A 1:1 volume ratio of lysozyme to agarose/trypsin  
was prepared by addition of 2  $\mu$ L of 500 nM lysozyme  
with 2  $\mu$ L of slurried agarose/trypsin (both prepared in  
a 20 mM phosphate; 10 mM DTT; pH 8.1 buffer). The  
20 slurried agarose contained approximately 1  $\mu$ L of beaded  
reagent. The combined volume was placed in a 600  $\mu$ L  
microcentrifuge tube and incubated at 40°C for 10  
minutes. The reaction was halted by the addition of 2  
 $\mu$ L of ACCA matrix and the total volume (with beaded  
25 reagent) of the digest/matrix mixture was placed on an  
inert mass spectrometer conventional sample  
presentation apparatus or probe tip and allowed to air  
dry.

Example 630 Time-of-Flight Mass Spectrometry

Mass spectrometry was performed using the sample  
presentation apparatus prepared in Example 3,  
Comparative Example 4, and Comparative Example 5 using  
a linear time-of-flight mass spectrometer. Desorption

and ionization were accomplished using the third harmonic (355 nm) output of the Nd:YAG laser (Continuum Surelite I). The optical path of the laser contained a variable attenuator (Newport 935), a right-angle beam steering prism and a 250 mm focal length quartz lens placed approximately 300 mm from the sample target. Ions generated from the matrix/analyte sample were accelerated to a potential of 30 kV using a two-stage acceleration source (total distance = 1.5 cm). An electrostatic particle guide was used to assist in ion transmission over the flight path. Detection was accomplished using a hybrid, microchannel-plate/discrete dynode, electron multiplier. Thresholds of ionization were determined empirically by increasing the intensity of the laser light at the sample surface while monitoring the ion signal real-time with a 50 Mhz oscilloscope (Tektronix TDS 310). Upon observation of stable ion signal, 50 laser shots were signal averaged using a 500 MS/s digital storage oscilloscope (Tektronix TDS 520A) and transferred to an IBM-compatible 486 computer. The data were analyzed using the PC compatible software LabCalc (Galactic Industries). All mass spectra were obtained in the positive-ion mode. External mass calibration was accomplished with the single and double-charged peaks of bovine insulin (MW = 5,733.5 Da). Masses of digest fragments were calculated using the known hen egg lysozyme amino acid sequence and the peptide data manipulation routine PROCOMP. This program is publicly available from Dr. Philip C. Andrews, University of Michigan Medical Center, Ann Arbor, Michigan 48104.

Prior to enzymatic cleavage, the integrity of the lysozyme samples was verified through MALDI analyses using sample presentation apparatus without any bound complex. No signals other than those due to the lysozyme ions, singly- and multiply-charged signals,



were observed during these analyses. Fig. 1 shows the results of a 10 minute digestion (40°C) of 10 picomoles of lysozyme using the sample presentation apparatus with trypsin bound to the gold surface as described in Example 3. No signals were observed over  $m/z = 6,000$ , which indicates high trypsin activity. Ion signals are observed for 33 out of 67 possible digest products (in the mass range up to 6,000 Da). Signals below  $m/z = 1,000$  Da were generally not considered because of possible interferences from the matrix. Signal intensity and quality was such to allow laser desorption conditions capable of resolving tryptic fragments at  $m/z = 1,424.8$ ,  $1,429.6$  and  $1,435.5$  Da (inset). Fig. 2 shows the correlation between the calculated masses of the enzymatically cleaved fragments and those observed in the mass spectrum.

Fig. 3 shows the mass spectrum obtained from using the same procedure performed with only 1 pmol of lysozyme applied to the surface. Virtually the same fragmentation pattern is observed, with an expected sacrifice in the signal-to-noise ratio of the data.

For comparison, 1 pmol of lysozyme was digested using either an equal mole ratio of free trypsin or an equal volume of agarose immobilized trypsin as described in Comparative Examples 4 and 5. Results of the digestion using free trypsin are shown in Fig. 4. Many of the low mass signals compliant with trypsin-generated fragments are observed which are shown in Fig. 6. Strong signals at  $m/z$  1177, 1459, 1498, 2166, 2191, and 2275, however, are attributed to autolysis products of the trypsin since these fragments are also observed during the MALDI mass analysis of trypsin as shown in Fig. 6.

Additionally, trypsin whole molecule signals (singly- and multiply- charged signals) also appear in the spectrum. The autolysis and parent signals

contaminate the data because these fragments are incorporated in the matrix. The trypsin used in digestion is not bound to the surface, and therefore, there is nothing to prevent the trypsin molecular ion and autolysis products from incorporating into the matrix crystals. In addition, a decrease in the signal-to-noise ratio is also observed, relative to Fig. 3, and is attributed to loss of sample in transfer and handling.

Agarose-immobilized trypsin was used in order to attempt to reduce the background contribution due to the trypsin autolysis products. The results obtained from the 10 minute, 40°C digestion of 1 pmole of lysozyme using approximately 1  $\mu$ L of beaded agarose as described in Comparative Example 5 are shown in Fig. 5. Signals are observed for nine lysozyme digest fragments, and only two autolysis product peaks at m/z 2164 and 3371 are found. In contrast to both the derivatized sample presentation apparatus and the free trypsin digestions, few peaks corresponding to digest fragments are observed. However, a marked decrease in the overall signal intensity is observed. Since the agarose is a porous medium, it is believed that digest fragments are lost through a combination of incorporation into the media, slow digestion due to poor localization of analyte around the immobilized enzyme, and inefficient elution using a small volume of matrix.

Comparing the spectrum obtained using the Au/trypsin derivatized sample presentation apparatus with those of the immobilized and free trypsin digests, it is clear that digests performed using the derivatized sample presentation apparatuses are noticeably free of background signals due to enzyme autolysis products and maintain a high level of activity. It is estimated from the digest pattern

which indicates the degree of total digestion that the activity of the derivatized surfaces in accordance with the invention falls somewhere between that of the agarose-immobilized and free trypsin (approximately 5 0.05 and 0.15 TAME units, respectively). This level of derivatized surface activity has been found to remain reasonably consistent from batch to batch if similar preparation conditions, e.g., incubation time, 10 temperature, and buffer, are used. However, activity has been found to be seriously impaired if the surfaces are inadvertently touched resulting in deposition of oils, or contaminated with adhesive during the fixing of the Au/trypsin foil to the conventional sample presentation apparatus. Care must be taken in the 15 final steps of surface preparation. Alternatively, the risk of contamination has been completely eliminated by sputter-coating the surfaces with gold and coupling trypsin directly to this gold coating. Trypsin activity using this method was found to be 20 approximately equal to that of the gold foil method.

#### Example 7

##### Au/trypsin Sample Presentation Surface Contacted with $\alpha$ -cobratoxin

$\alpha$ -cobratoxin was analyzed using an immobilized 25 trypsin sample presentation apparatus as described in Example 2.

First, the unfragmented  $\alpha$ -cobratoxin was analyzed using the MALDI method. Signals were obtained which correspond the singly and doubly charged ions of the 30 intact peptide. As can be seen from Fig. 7 and Fig. 8, singly and doubly charged ions corresponding to residue 1-71 are observed at 7,823 and 3,912 Da, respectfully. Signals less than m/z approximately 1,000 Da are due to the ACCA matrix.

Next 10 picomoles of  $\alpha$ -cobratoxin were digested for 30 minutes with a trypsin surface prepared according to Example 2. The  $\alpha$ -cobratoxin was applied to the surface in a 3  $\mu$ L volume of 10 mM phosphate buffer (pH 7.8). The reaction was allowed to proceed in a highly humid environment, at room temperature, for 30 minutes before being stopped by the addition of an equal volume of ACCA matrix. The mixture was allowed to dry before being inserted into the mass spectrometer. The resulting mass spectrum, shown in Fig. 9, shows singly and doubly charged ion signals for peptides with molecular weights of 7,445, 7,575 and 7,823 Da. These weights correspond to peptide sequences 1-68, 1-69 and 1-71, respectively, as shown in Fig. 7. These are the primary fragments expected from the tryptic treatment of the non-reduced peptide. Five other cleavage sites are possible, each producing different molecular weight fragments. However, these sites are sterically hindered and not available to trypsin. Therefore, these five other possible peaks were not observed.

$\alpha$ -cobratoxin was further analyzed by repeating the above-described 10 picomole  $\alpha$ -cobratoxin digestion by surface-bound trypsin in the presence of a reducing agent (10 pmol  $\alpha$ -cobratoxin in 3  $\mu$ L - 10 mM phosphate; 10 mM DTT). As can be seen from Fig. 7, the reducing agent broke the disulfide bonds, thus enabling the generation of more fragments. The resulting spectrum shown in Figs. 10 and 11 shows a set of peptide fragments indicative of reduced cobratoxin.

#### Example 8

##### Preparation of Sample Presentation Apparatus with Surface-Bound $\alpha$ -Chymotrypsin

A mass spectrometry sample presentation apparatus was prepared by binding  $\alpha$ -chymotrypsin to the surface

of the apparatus as follows.  $\alpha$ -chymotrypsin or TPCK-treated  $\alpha$ -chymotrypsin was dissolved in 20 mM phosphate buffer, pH 7.8 to a concentration of 1 mg/mL for use in preparation of a derivatized sample presentation apparatus as described in Example 2.

#### Example 9

##### Sequence Determination For Design of Nucleic Acid Probes And Gene Identification

The combination of surface-immobilized and free-peptidases were used to generate heterogeneous fragment mixtures from an original peptide as follows.

Enzymatically active trypsin and  $\alpha$ -chymotrypsin mass spectrometry sample presentation surfaces were prepared as described in Examples 2 and 8.

The synthetic peptide human parathyroid hormone 1-34 (hPTH) was chosen as a test peptide for analyses. The sequence of the peptide is:

10	20	30
SVSEIQLMHN LGKHLNSMER VEWLRRKKLQD VHNF		

Digestions were performed in the following manner. Three microliter aliquots of a 0.01 mg/mL solution of hPTH (in a buffer of 20 mM NaHPO<sub>4</sub>, pH 8.1) were applied to derivatized mass spectrometer sample presentation apparatus prepared in accordance with Example 2 or Example 8. The digestion was carried out in a high humidity environment at a temperature of between 40 and 50°C. The digestions were allowed to run for 5 to 10 minutes before 0.5 -1.0  $\mu$ L aliquots of the reaction mixtures were removed from the surfaces. These aliquots were MALDI analyzed for endopeptidase mapping. The pH was reduced to approximately 5 with the addition of an aliquot of 20 mM NH<sub>4</sub>COOCH<sub>3</sub>, the volume of which

was equal to that of the remaining reaction mixture. 500 nL aliquots of 0.001 mg/mL carboxypeptidase P (CPP), in the acetate buffer, were then added to the reaction mixtures. Protein ladder digestions were  
5 allowed to run for between 1 to 3 minutes before the surfaces were removed from the heated high humidity environment and the reactions halted through the addition of an equal volume of MALDI matrix. The matrix/reaction mixture was allowed to air-dry and the  
10 enzymatically-active surfaces inserted directly into the mass spectrometer.

Fig. 12 is an ACCA-MALDI mass spectrum of 2.5 picomoles of the hPTH, the starting point of reference for the following enzymatic analyses. A strong singly-  
15 protonated signal is observed at a mass-to-charge ratio ( $m/z$ ) of 4,119.0 Da, a doubly-charged at  $m/z = 2,019.7$  Da, and little else other than matrix ion signals at  $m/z < 500$  Da.

The hPTH was further analyzed after a 10 minute  
20 digestion on an Au/ $\alpha$ -chymotrypsin surface maintained at 40°C, pH 8.1. The results of this digestion are shown in Fig. 13. Four major signals at 1,398.5, 1,706.2, 1850.0 and 2,738.7 Da are observed as shown in Fig. 13.

Before adding the MALDI matrix to obtain the mass  
25 spectrum shown in Fig. 13, a portion of the digested peptide was transferred to a second conventional sample presentation apparatus surface. The portion of the digested peptide on the second conventional sample presentation surface was adjusted to a pH of  
30 approximately 5 and further exposed to cpp. Fig. 14 shows the MALDI mass spectrum obtained from this second sample. A series of ion signals due to cpp digestion is observed in the  $m/z$  range of 1,800 - 2,800 Da, presumably stemming from the Au/ $\alpha$ -chymotrypsin digest  
35 signal at  $m/z = 2,738.7$ . The first residue is a preferred cleavage site for  $\alpha$ -chymotrypsin. Fig. 15

summarizes the masses of the signals observed in Fig. 14, the differences in mass between adjacent signals and the residues determined from the differences. The partial sequence (read as shown in Fig. 15 from N to C-terminal is determined by sequential mass difference to give (pep) - MERVEW - COOH.

A second similar analysis of the hPTH was performed using an Au/trypsin sample presentation surface followed by cpp digestion. Fig. 16 shows the results of the Au/trypsin digestion after 5 minutes at a temperature of 45°C (pH 8.1). Fig. 18 summarizes the masses observed for the digestion. The trypsin digestion shown in Fig. 16 resulted in the production of several starting fragments, and indicated the presence of the short sequence (Arg/Lys) - Lys - Lys.

Before adding the MALDI matrix to obtain the mass spectrum shown in Fig. 16, a portion of the digested peptide was transferred to a second, conventional sample presentation apparatus surface. The cpp was then added to this second portion. After pH adjustment to approximately 5, the ladder digestion was allowed to proceed for 3 minutes before being terminated by the addition of ACCA to the sample presentation surface. Fig. 17 shows the resulting sequence ladder. From the observed mass differences summarized in Fig. 18 the sequence was determined to be (pep) - NSMERVEW(L/I)RK - COOH. The first residue cleaved by cpp is that of Arg. Arg is a preferred cleavage site for trypsin. This is obviously an extension of the partial sequence determined using Au/ $\alpha$ -chymotrypsin/cpp.

The large scale sequence, (Arg/Lys) - Lys -Lys, overlaps with the ladder sequence at correct mass. This allows both spectra to be combined for sequence extension by an additional Lys. The first residue observed in the Au/ $\alpha$ -chymotrypsin/cpp series, from the C-terminal end, was determined to be tryptophan. This

is expected because tryptophan is one of the preferred sites of  $\alpha$ -chymotrypsin cleavage.

There are two ion signals to chose from in the region of the last cleavage of the Au/ $\alpha$ -  
5 chymotrypsin/cpp digestion as can be seen from Fig. 15, inset, leading to the possibility of an ambiguity. The two mass differences represent the possibility of either a serine or a proline as the final residue. On the basis of the mass assignment error observed in the  
10 entire series (average approximately 0.15 Da), the final determination of serine, with -0.1 Da error was made over that of proline (with -2.2 Da error). The resulting sequence was determined to be (pep) - SMERVEV - COOH.

15 A nominal mass difference of 128 Da was observed at several points in the spectrum resulting from the initial trypsin cleavage in the Au/trypsin/cpp analysis. Considering that trypsin will cleave only at lysine or arginine, the 128 Da mass difference is  
20 highly likely to result from a lysine C-terminal to either another lysine or an arginine. Likewise, two adjacent mass differences of 128 Da, as observed in the 3,000 to 3,300 Da region, indicates the presence of Lys-Lys in the sequence, with the third residue being  
25 either another lysine or an arginine. This information, in combination with recognizing overlaps in the endo- and exopeptidase spectra, allows the two spectra to be used in combination to extend the sequence determination. Fig. 19 illustrates a ladder  
30 sequence extended by an additional lysine determined from the (Arg/Lys) - Lys-Lys observed in the trypsin spectrum overlapping with the Arg-Lys of the ladder sequence. The 128 Da mass differences observed in this manner can be unambiguously assigned to Lys as opposed  
35 to Gln which has the same molecular weight. The result



is that 12 residues of the sequence were determined to be (pep) -NSMERVIEW(L/I)RKK - COOH.

In gathering this data, four 50 to 100 laser-shot mass spectra, with quality and signal-to-noise great  
5 enough to determine relative molecular masses within approximately 0.2 Da, were generated using a total of approximately 15 picomoles of sample. Typical of the MALDI process, the sample preparations were capable of thousands of single laser-shot mass spectra of equal  
10 quality. As no special efforts were taken to minimize the amount of sample applied to the sample presentation surface apparatus, and the surface area of the apparatus was quite large, it may be possible to reduce the analyte sample amount, without sacrificing quality,  
15 by simply downscaling the whole process by reduction of surface area and application of less analyte sample.

It was important to maintain the derivatized sample presentation surfaces at elevated temperatures in order to perform these digestions rapidly. The  
20 temperatures used for both the endo- and exopeptidase digestions, performed at elevated temperatures of up to approximately 55°C, are somewhat above that typically used in proteolytic digestions, namely, 37°C. This, combined with the high relative activity of the  
25 enzymatically-active sample presentation surfaces, is believed to be responsible for the rapid analysis times. Although not guaranteed in every case, the 15 minute analysis time demonstrated in hPTH example represents a rate of analysis of just under one residue  
30 per minute.

After a peptide sequence is determined by the above-described process, the rules of degeneracy can be used to identify many nucleic acid sequences which code for the peptide sequence. See, e.g., Biochemistry,  
35 Stryer, pp. 91-115 (1988). Since many different nucleic acid sequences code for a single peptide,

several nucleic acid sequences can be generated from the peptide sequences. The nucleic acid sequences can be used to search a database of nucleic acid sequences, e.g., Genbank, to identify genes which code for a particular polypeptide.

#### Example 10

##### Preparation of Sample Presentation Apparatus with Surface-Bound Pepsin

A mass spectrometry sample presentation apparatus was prepared by binding pepsin to the surface of the apparatus as follows. Gold foil was first activated with DSP according to the method given in Example 1. The DSP activated gold was then reacted with ethylene diamine (20% v/v in isopropanol) at room temperature for 30 minutes. After incubation, the gold foil was rinsed thoroughly with isopropanol and then reacted with 0.1 M[1-ethyl-3-(3-dimethylaminopropyl)carbodiimide] in 20 mM phosphate buffer at pH 4.5, for 30 minutes at room temperature. The foil was then rinsed thoroughly with the phosphate buffer. Next, a solution of 10 mg/ml pepsin in phosphate buffer was added to the surface and allowed to incubate for 4 hours at 4°C. After incubation the foil was rinsed repeatedly with phosphate buffer and finally vacuum dried.

#### Example 11

##### Au/pepsin Sample Presentation Surface Contacted with Pepsin

A digestion was performed with the Au/Pepsin active surfaces prepared in Example 10 by the application of approximately 5 pmole of hPTH 1-34 in 10 mM ammonium acetate buffer (pH 3.4). The hPTH was digested for 8 minutes at 45°C before stopping the reaction by adding ACCA matrix. Time-of-Flight mass

spectrometry was then performed. The resulting spectrum shown in Fig. 20 includes ion signals for the pepsin-generated peptides corresponding to amino acids 4-18, 9-24, 8-24, and 18-34.

5

#### Example 12

##### Sequential Treatment of a Biomolecule By Two Different Complexes

The combination of two different surface bound complexes was used to modify a biomolecule in sequence as follows. First hPTH was digested for 10 minutes at 35°C, pH 8.1 using an immobilized Au/ $\alpha$ -chymotrypsin sample presentation device as described in Example 2. A portion of the digest mixture was then transferred to a cpp activated apparatus prepared as described in Example 2 and the pH of the solution was adjusted to approximately 5. The cpp digestion was allowed to proceed for 10 minutes at 35°C before termination by the addition of the ACCA matrix.

The mass spectrum of the results of the initial Au/ $\alpha$ -chymotrypsin digestion is shown in Fig. 21. Ion signals resulting from the  $\alpha$ -chymotrypsin cleavage of hPTH are observed. The mass spectrum of the results of the Au/cpp digestion of the  $\alpha$ -chymotrypsin digestion mixture is shown in Fig. 22. New ion signals indicated by an asterisk "\*" are observed. These new signals correspond to the C-terminal digestion of the original peptide species.

#### Example 13

##### Purification of Biomolecules

The invention can be used to purify biomolecules as follows. For many biological studies, it is essential to remove undesirable proteases from biological solutions. Aprotinin is a single-chain basic 58-amino-acid polypeptide with a molecular mass

of 6512 Daltons that has a polyvalent inhibitory action on proteases. Specifically, it inhibits a variety of enzymes belonging to the family of serine proteases by binding to the active site of the enzyme, forming tight  
5 complexes. A mass spectrometry sample presentation apparatus is prepared by tethering aprotinin to the surface in a manner similar to that described in Example 2. A solution of a polypeptide and a protease is added to the surface. The protease is removed from  
10 solution by the aprotinin. Mass spectrometry is performed and only the purified polypeptide should be observed.

#### Example 14

##### Preparation of A Sample Presentation Apparatus 15 Which Purifies Biomolecules By Ion Exchange

An ion exchange functionality such as a diethylaminoethyl derivative, a t-butyl amino derivative (for anion exchange), a sulfite derivative, a carboxylic acid derivative (for cation exchange), or  
20 a combination of the above is immobilized to a sample presentation surface prepared according to Example 1.

#### Example 15

##### Purification of Biomolecules By Ion Exchange

25 A biomolecule-containing solution is added to a surface prepared according to Example 14. The ion exchange functionality bound to the sample presentation surface exchanges ions with detrimental ionic species present as impurities in the biomolecule-containing  
30 solution.

Example 16Analysis of DNA Using  
Sample Presentation Apparatus

First, the nuclease, snake-venom

- 5 phosphodiesterase, is attached to the sample  
presentation apparatus surface. This attachment may be  
accomplished using a tethering molecule in a manner  
similar to that described in Example 2. Second, DNA is  
added to the surface. After the phosphodiesterase has  
10 produced DNA fragments, a MALDI matrix is added, and  
MALDI time-of-flight mass spectrometry is performed.

Example 17Analysis of Carbohydrate Using  
Sample Presentation Apparatus

- 15 An enzyme such as mannanase which digests  
carbohydrates is attached to the sample presentation  
apparatus surface. This attachment may be accomplished  
using a tethering molecule in a manner similar to that  
described in Example 2. Next, a carbohydrate is added  
20 to the surface. After the mannanase has produced  
carbohydrate fragments, a MALDI matrix is added, and  
MALDI time-of-flight mass spectrometry is performed.

- Many other variations and modifications of the  
invention will be apparent to those skilled in the art  
25 without departing from the spirit and scope of the  
invention. The above-described embodiments are,  
therefore intended to be merely exemplary, and all such  
variations and modifications are intended to be  
included within the scope of the claims.

Claims

- 1 1. A sample presentation apparatus comprising:  
2 (a) a mass spectrometry sample  
3 presentation surface; and  
4 (b) a complex bound to the surface, the  
5 complex including at least one molecule which  
6 chemically modifies a biomolecule.
- 1 2. The sample presentation apparatus according to  
2 claim 1, wherein the biomolecule comprises at  
3 least one selected from the group consisting of  
4 polypeptides, DNA, carbohydrates, RNA and  
5 combinations thereof.
- 1 3. A sample presentation apparatus according to  
2 claim 2, wherein the complex comprises a tethering  
3 molecule which is attached to the surface, and a  
4 reactive molecule which chemically modifies a  
5 biomolecule; wherein the tethering molecule is  
6 attached to the reactive molecule.
- 1 4. A sample presentation apparatus according to  
2 claim 2, wherein the complex purifies the  
3 biomolecule.
- 1 5. A sample presentation apparatus according to  
2 claim 2, wherein the complex selectively digests  
3 the biomolecule at specific locations.
- 1 6. A sample presentation apparatus according to  
2 claim 2, wherein the biomolecule is a polypeptide,  
3 and the complex comprises a proteolytic enzyme.
- 1 7. A sample presentation apparatus according to  
2 claim 6, wherein the proteolytic enzyme is

- 3       selected from the group consisting of trypsin,  $\alpha$ -  
4       chymotrypsin, pepsin and carboxypeptidase-P.
- 1   8.   A method of making a sample presentation  
2       apparatus, which comprises providing a mass  
3       spectrometry sample presentation apparatus  
4       surface; and binding a complex to the surface  
5       wherein the complex includes at least one molecule  
6       which chemically modifies a biomolecule.
- 1   9.   A method of making a sample presentation apparatus  
2       according to claim 8, wherein the step of binding  
3       a complex to the sample presentation surface  
4       comprises:  
5               (a) binding at least one end of a  
6       tethering molecule to the mass spectrometry sample  
7       presentation surface; and  
8               (b) binding a reactive molecule to the  
9       tethering molecule wherein the reactive molecule  
10      chemically modifies a biomolecule.
- 1   10.  A method of analyzing a biomolecule comprising the  
2       steps of:  
3               (a) providing a mass spectrometry  
4       presentation surface,  
5               (b) binding at least one complex  
6       molecule to the surface, the complex including at  
7       least one molecule which chemically modifies a  
8       biomolecule;  
9               (c) contacting the biomolecule with the  
10      surface, thereby chemically modifying the  
11      biomolecule;  
12              (d) applying a MALDI matrix to the  
13      surface; and

14 (e) determining the molecular weight of  
15 the chemically modified biomolecule in a mass  
16 spectrometer.

1 11. A method of analyzing a biomolecule according to  
2 claim 10, wherein the biomolecule comprises at  
3 least one selected from the group consisting of  
4 polypeptides, DNA, carbohydrates, RNA and  
5 combinations thereof.

1 12. A method of analyzing a biomolecule according to  
2 claim 11, wherein the complex comprises a  
3 tethering molecule which is attached to the  
4 surface at at least one end of the tethering  
5 molecule and a reactive molecule which chemically  
6 modifies a biomolecule, wherein the tethering  
7 molecule is attached to the reactive molecule.

1 13. A method of analyzing a biomolecule according to  
2 claim 11, wherein the complex purifies a  
3 biomolecule.

1 14. A method of analyzing a biomolecule according to  
2 claim 11, wherein the complex selectively digests  
3 the biomolecule at specific locations.

1 15. A method of analyzing a biomolecule according to  
2 claim 11, wherein the biomolecule is a polypeptide  
3 and the complex comprises a proteolytic enzyme.

1 16. A method of analyzing a polypeptide according to  
2 claim 15, wherein the proteolytic enzyme is  
3 selected from the group consisting of trypsin,  $\alpha$ -  
4 chymotrypsin, pepsin and carboxypeptidase-P.



- 1    17.    A method of analyzing a polypeptide according to  
2            claim 15 wherein a reducing agent which cleaves  
3            disulfide bonds is added to the surface.
- 1    18.    A method of analyzing a polypeptide according to  
2            claim 15 wherein a reducing agent which cleaves  
3            disulfide bonds is attached to the surface.
- 1    19.    A method of analyzing a biomolecule according to  
2            claim 11, wherein after the step of contacting the  
3            biomolecule with the surface, the surface and the  
4            biomolecule are incubated in a moist atmosphere.
- 1    20.    A method of analyzing a biomolecule according to  
2            claim 11, wherein after the biomolecule is  
3            modified by the complex, the biomolecule is  
4            exposed to a second reagent molecule which  
5            modifies the biomolecule a second time.
- 1    21.    A method of analyzing a biomolecule according to  
2            claim 20, wherein the second reagent is different  
3            from the reagent molecule in the complex.
- 1    22.    A method of analyzing a biomolecule according to  
2            claim 21, wherein the second reagent is a complex  
3            which is bound to a surface.
- 1    23.    A method of analyzing a biomolecule according to  
2            claim 20, wherein the molecular weight data  
3            obtained from analyzing the biomolecules modified  
4            by the complex, and the molecular weight data  
5            obtained from the biomolecules modified by both  
6            the complex and the second reagent, is compared to  
7            determine the partial structure of the  
8            biomolecule.

- 1 24. A method of analyzing a biomolecule according to  
2 claim 23, wherein the molecular weight data is  
3 used to determine a limited amino acid sequence of  
4 a peptide.
- 1 25. A method according to claim 24 further comprising  
2 determining at least one nucleic acid sequence  
3 corresponding to the limited amino acid sequence,  
4 and searching a database containing nucleic acid  
5 sequences for the nucleic acid sequence.
- 1 26. A method of analyzing a biomolecule according to  
2 claim 11, wherein the mass spectrometer comprises:  
3 (a) a mass spectrometry sample  
4 presentation apparatus which comprises said mass  
5 spectrometry presentation surface:  
6 (b) means for introducing the sample  
7 presentation surface into a volatilization  
8 chamber;  
9 (c) means for volatilizing and ionizing  
10 the chemically modified biomolecule; and  
11 (d) means for determining the molecular  
12 weight to charge ratio of the chemically modified  
13 biomolecule ion based on its trajectory in the  
14 mass spectrometer.
- 1 27. A method of analyzing a biomolecule according to  
2 claim 26, wherein the mass spectrometer further  
3 comprises means for generating an electric field,  
4 the field being oriented to cause the chemically  
5 modified biomolecule ions to travel away from  
6 the sample presentation apparatus and toward a  
7 detecting means.
- 1 28. A method of analyzing a biomolecule according to  
2 claim 26, wherein the means for volatilizing and

3       ionizing the chemically modified biomolecule is a  
4       LASER.

1   29. A method of analyzing a biomolecule according to  
2       claim 26, wherein the means for determining the  
3       molecular weight of the chemically modified  
4       biomolecule measures the time of flight for the  
5       chemically modified biomolecule.

1   30. A mass spectrometer, which comprises:  
2       (a) a mass spectrometry sample  
3       presentation apparatus which comprises:  
4           (i) a mass spectrometry sample  
5       presentation surface; and  
6           (ii) a complex bound to the  
7       surface, the complex including at least one molecule  
8       which chemically modifies a biomolecule;  
9       (b) means for introducing the sample  
10      presentation surface into a volatilization  
11      chamber;  
12      (c) means for volatilizing and ionizing  
13      the chemically modified biomolecule; and  
14      (d) means for determining the molecular  
15      weight to charge ratio of the chemically modified  
16      biomolecule ion based on its trajectory.

1   31. A mass spectrometer according to claim 30, wherein  
2       the mass spectrometer further comprises means for  
3       generating an electric field, the field being  
4       oriented to cause the chemically modified  
5       biomolecule ions to travel away from the sample  
6       presentation apparatus and toward a detecting  
7       means.

1   32. A mass spectrometer according to claim 30, wherein  
2       the biomolecule comprises at least one selected

- 3 from the group consisting of polypeptides, DNA,  
4 carbohydrates, RNA and combinations thereof.
- 1 33. A mass spectrometer according to claim 32, wherein  
2 the complex comprises a tethering molecule which  
3 is attached at at least one end to the surface,  
4 and a reactive molecule which chemically modifies  
5 a biomolecule wherein the tethering molecule is  
6 attached to the reactive molecule.
- 1 34. A mass spectrometer according to claim 32, wherein  
2 the complex purifies a biomolecule.
- 1 35. A mass spectrometer according to claim 32, wherein  
2 the complex selectively digests the biomolecule at  
3 specific locations in the biomolecule.
- 1 36. A mass spectrometer according to claim 32, wherein  
2 the biomolecule is a polypeptide, and the reactive  
3 molecule is a proteolytic enzyme.
- 1 37. A mass spectrometer according to claim 36, wherein  
2 the proteolytic enzyme is selected from the group  
3 consisting of trypsin,  $\alpha$ -chymotrypsin, pepsin and  
4 carboxypeptidase-P.
- 1 38. A mass spectrometer according to claim 30, wherein  
2 the means for volatilizing and ionizing the  
3 chemically modified biomolecule is a LASER.
- 1 39. A mass spectrometer according to claim 38, wherein  
2 the means for determining the molecular weight of  
3 the chemically modified biomolecule measures the  
4 time of flight for the chemically modified  
5 biomolecule.

- 1 40. A method of analyzing a polypeptide according to  
2 claim 15, wherein after the molecular weight of a  
3 polypeptide fragment is determined, the molecular  
4 weight is subtracted from the weight of a next  
5 lower molecular weight polypeptide fragment to  
6 determine which amino acid is lacking from the  
7 lower molecular weight polypeptide fragment.
- 1 41. A method of analyzing a polypeptide according to  
2 claim 40, wherein a limited sequence of a  
3 polypeptide, containing amino acids, is determined  
4 by sequentially analyzing lower molecular weight  
5 polypeptide fragments to determine which amino  
6 acids they lack, and further comprising  
7 determining at least one nucleic acid sequence  
8 corresponding to the limited amino acid sequent  
9 and searching a database containing nucleic acid  
10 sequences for the nucleic acid sequence.

## AMENDED CLAIMS

[received by the International Bureau on 27 November 1996 (27.11.96);  
original claims 1-41 replaced by amended claims 1-43 (9 pages)]

Claims

- 1 1. A sample presentation apparatus comprising:  
2 (a) a mass spectrometry sample  
3 presentation apparatus having a mass spectrometry  
4 sample presentation surface; and  
5 (b) a complex immobilized on the  
6 surface, the complex including at least one  
7 molecule which chemically modifies a biomolecule,  
8 wherein the biomolecule is selected from the group  
9 consisting of polypeptides, DNA, carbohydrates,  
10 RNA, and combinations thereof.
- 1 2. A sample presentation apparatus according to claim  
2 1, wherein the complex comprises:  
3 (a) a tethering molecule which is  
4 attached to the sample presentation surface at at  
5 least one end of the tethering molecule, and  
6 (b) a reactive molecule which chemically  
7 modifies a biomolecule;  
8 wherein the tethering molecule is  
9 attached to the reactive molecule.
- 1 4. A sample presentation apparatus according to  
2 claim 1, wherein the complex purifies the  
3 biomolecule.
- 1 5. A sample presentation apparatus according to  
2 claim 1, wherein the complex selectively digests  
3 the biomolecule at specific locations.
- 1 6. A sample presentation apparatus according to  
2 claim 3, wherein the biomolecule is a polypeptide,  
3 and the reactive molecule is a proteolytic enzyme.
- 1 7. A sample presentation apparatus according to  
2 claim 6, wherein the proteolytic enzyme is

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- 3 selected from the group consisting of trypsin,  $\alpha$ -  
4 chymotrypsin, pepsin and carboxypeptidase-2.
- 1 8. A method of making a sample presentation  
2 apparatus, which comprises providing a mass  
3 spectrometry sample presentation apparatus having  
4 a mass spectrometry sample presentation surface;  
5 and binding a complex to the surface wherein the  
6 complex includes at least one molecule which  
7 chemically modifies a biomolecule, wherein the  
8 biomolecule is selected from the group consisting  
9 of polypeptides, DNA, carbohydrates, RNA, and  
10 combinations thereof.
- 1 9. A method of making a sample presentation apparatus  
2 according to claim 8, wherein the step of binding  
3 a complex to the sample presentation surface  
4 comprises:  
5 (a) binding at least one end of a  
6 tethering molecule to the mass spectrometry sample  
7 presentation surface; and  
8 (b) binding a reactive molecule to the  
9 tethering molecule wherein the reactive molecule  
10 chemically modifies a biomolecule, wherein the  
11 biomolecule is selected from the group consisting  
12 of polypeptides, DNA, carbohydrates, RNA, and  
13 combinations thereof.
- 1 10. A method of analyzing a biomolecule comprising the  
2 steps of:  
3 (a) providing a mass spectrometry  
4 sample presentation surface;  
5 (b) contacting at least one complex  
6 including at least one molecule which chemically  
7 modifies a biomolecule with the sample

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8 presentation surface to form a surface-bound  
9 complex;  
10 (c) removing any unbound complex from  
11 the sample presentation surface;  
12 (d) contacting the biomolecule with the  
13 surface-bound complex, thereby chemically  
14 modifying the biomolecule;  
15 (e) applying a MALDI matrix to the  
16 sample presentation surface; and  
17 (f) determining the molecular weight of  
18 the chemically modified biomolecule in a mass  
19 spectrometer.

1 11. A method of analyzing a biomolecule according to  
2 claim 10, wherein the biomolecule comprises at  
3 least one selected from the group consisting of  
4 polypeptides, DNA, carbohydrates, RNA and  
5 combinations thereof.

1 12. A method of analyzing a biomolecule according to  
2 claim 11, wherein the complex comprises:  
3 (a) a tethering molecule which is  
4 attached to the sample presentation surface, at at  
5 least one end of the tethering molecule, and  
6 (b) a reactive molecule which  
7 chemically modifies a biomolecule,  
8 wherein the tethering molecule is also  
9 attached to the reactive molecule.

1 13. A method of analyzing a biomolecule according to  
2 claim 11, wherein the complex purifies a  
3 biomolecule.

1 14. A method of analyzing a biomolecule according to  
2 claim 11, wherein the complex selectively digests  
3 the biomolecule at specific locations.



- 1 15. A method of analyzing a biomolecule according to  
2 claim 11, wherein the biomolecule is a polypeptide  
3 and the complex comprises a proteolytic enzyme.
- 1 16. A method of analyzing a polypeptide according to  
2 claim 15, wherein the proteolytic enzyme is  
3 selected from the group consisting of trypsin,  $\alpha$ -  
4 chymotrypsin, pepsin and carboxypeptidase-P.
- 1 17. A method of analyzing a polypeptide according to  
2 claim 15 wherein a reducing agent which cleaves  
3 disulfide bonds is added to the surface.
- 1 18. A method of analyzing a polypeptide according to  
2 claim 15 wherein a reducing agent which cleaves  
3 disulfide bonds is attached to the surface.
- 1 19. A method of analyzing a biomolecule according to  
2 claim 11, wherein after the step of contacting the  
3 biomolecule with the surface-bound complex, the  
4 surface-bound complex and the biomolecule are  
5 incubated in a moist atmosphere.
- 1 20. A method of analyzing a biomolecule according to  
2 claim 11, wherein after the biomolecule is  
3 modified by the complex, the biomolecule is  
4 exposed to a second reagent molecule which  
5 modifies the biomolecule a second time.
- 1 21. A method of analyzing a biomolecule according to  
2 claim 20, wherein the second reagent molecule is  
3 different from the reactive molecule.
- 1 22. A method of analyzing a biomolecule according to  
2 claim 21, wherein the second reagent molecule is a  
3 complex which is bound to a surface.

- 1 23. A method of analyzing a biomolecule according to  
2 claim 20, wherein the molecular weight data  
3 obtained from analyzing the biomolecules modified  
4 by the complex, and the molecular weight data  
5 obtained from the biomolecules modified by both  
6 the complex and the second reagent molecule, is  
7 compared to determine the partial structure of the  
8 biomolecule.
- 1 24. A method of analyzing a biomolecule according to  
2 claim 23, wherein the molecular weight data is  
3 used to determine a limited amino acid sequence of  
4 a peptide.
- 1 25. A method according to claim 24 further comprising  
2 determining at least one nucleic acid sequence  
3 corresponding to the limited amino acid sequence,  
4 and searching a database containing nucleic acid  
5 sequences for the nucleic acid sequence.
- 1 26. A method of analyzing a biomolecule according to  
2 claim 11, wherein the mass spectrometer comprises:  
3 (a) a mass spectrometry sample  
4 presentation apparatus which comprises said mass  
5 spectrometry presentation surface:  
6 (b) means for introducing the sample  
7 presentation surface into a volatilization  
8 chamber;  
9 (c) means for volatilizing and ionizing  
10 the chemically modified biomolecule; and  
11 (d) means for determining the molecular  
12 weight to charge ratio of the chemically modified  
13 biomolecule ion based on its trajectory in the  
14 mass spectrometer.

- 1 27. A method of analyzing a biomolecule according to  
2 claim 26, wherein the mass spectrometer further  
3 comprises means for generating an electric field,  
4 the field being oriented to cause the chemically  
5 modified biomolecule ions to travel away from  
6 the sample presentation surface and toward a  
7 detecting means.
- 1 28. A method of analyzing a biomolecule according to  
2 claim 26, wherein the means for volatilizing and  
3 ionizing the chemically modified biomolecule is a  
4 LASER.
- 1 29. A method of analyzing a biomolecule according to  
2 claim 26, wherein the means for determining the  
3 molecular weight of the chemically modified  
4 biomolecule measures the time of flight for the  
5 chemically modified biomolecule.
- 1 31. A mass spectrometer according to claim 43, wherein  
2 the mass spectrometer further comprises means for  
3 generating an electric field, the field being  
4 oriented to cause the chemically modified  
5 biomolecule ions to travel away from the sample  
6 presentation apparatus and toward a detecting  
7 means.
- 1 32. A mass spectrometer according to claim 43, wherein  
2 the biomolecule comprises at least one selected  
3 from the group consisting of polypeptides, DNA,  
4 carbohydrates, RNA and combinations thereof.
- 1 33. A mass spectrometer according to claim 32, wherein  
2 the complex comprises a tethering molecule which  
3 is attached at at least one end to the surface,  
4 and a reactive molecule which chemically modifies

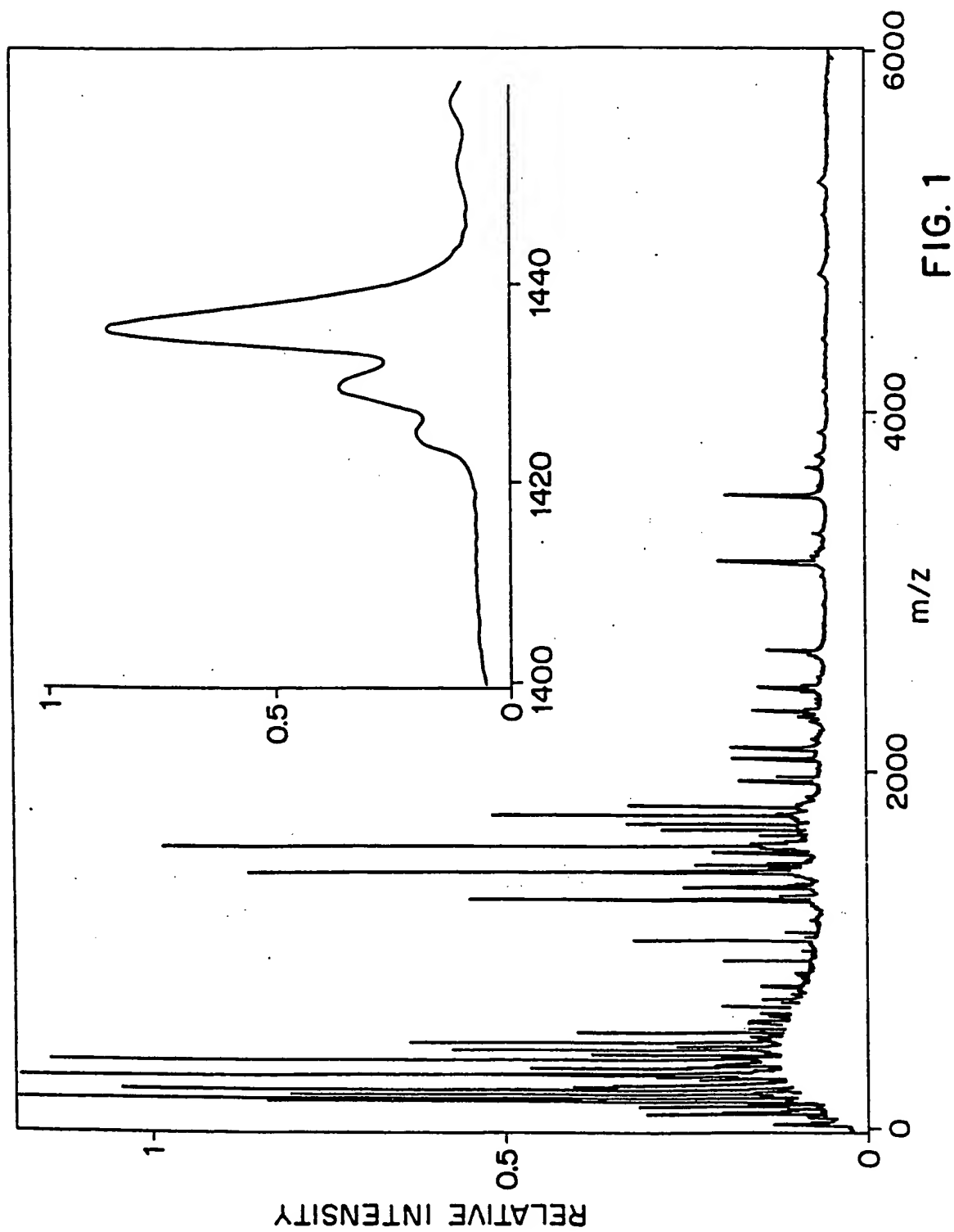
-50-

- 5 a biomolecule wherein the tethering molecule is  
6 attached to the reactive molecule.
- 1 34. A mass spectrometer according to claim 32, wherein  
2 the complex purifies a biomolecule.
- 1 35. A mass spectrometer according to claim 32, wherein  
2 the complex selectively digests the biomolecule at  
3 specific locations in the biomolecule.
- 1 36. A mass spectrometer according to claim 33, wherein  
2 the biomolecule is a polypeptide, and the reactive  
3 molecule is a proteolytic enzyme.
- 1 37. A mass spectrometer according to claim 36, wherein  
2 the proteolytic enzyme is selected from the group  
3 consisting of trypsin,  $\alpha$ -chymotrypsin, pepsin and  
4 carboxypeptidase-P.
- 1 38. A mass spectrometer according to claim 43, wherein  
2 the means for volatilizing the matrix material is  
3 a LASER.
- 1 39. A mass spectrometer according to claim 38, wherein  
2 the means for determining the molecular weight to  
3 charge ratio of the chemically modified  
4 biomolecule measures the time of flight for the  
5 chemically modified biomolecule.
- 1 40. A method of analyzing a polypeptide according to  
2 claim 15, wherein after the molecular weight of a  
3 polypeptide fragment is determined, the molecular  
4 weight is subtracted from the weight of a next  
5 lower molecular weight polypeptide fragment to  
6 determine which amino acid is lacking from the  
7 lower molecular weight polypeptide fragment.

- 1 41. A method of analyzing a polypeptide according to  
2 claim 40, wherein a limited sequence of a  
3 polypeptide containing amino acids, is determined  
4 by sequentially analyzing lower molecular weight  
5 polypeptide fragments to determine which amino  
6 acids they lack, and further comprising  
7 determining at least one nucleic acid sequence  
8 corresponding to the limited amino acid sequence  
9 and searching a database containing nucleic acid  
10 sequences for the nucleic acid sequence.
- 1 42. A mass spectrometer, which comprises a mass  
2 spectrometry sample presentation apparatus which  
3 comprises:  
4 (a) a mass spectrometry sample  
5 presentation surface; and  
6 (b) a complex immobilized on the  
7 surface, the complex including at least one  
8 molecule which chemically modifies a biomolecule  
9 wherein the complex is not incorporated in a MALDI  
10 matrix and is not volatilized in a mass  
11 spectrometer.
- 1 43. A mass spectrometer according to claim 42, which  
2 further comprises:  
3 (a) means for introducing a sample  
4 presentation surface into a volatilization  
5 chamber;  
6 (b) means for volatilizing and ionizing  
7 a chemically modified biomolecule, thereby forming  
8 a chemically modified biomolecule ion, said means  
9 for volatilizing comprising (i) a matrix material  
10 which solubilizes the chemically modified  
11 biomolecule and absorbs light energy at a  
12 frequency easily accessible by a laser, wherein  
13 the matrix material is applied over the complex,

14 and (ii) means for volatilizing the matrix  
15 material; and  
16 (c) means for determining a molecular  
17 weight to charge ratio of the chemically modified  
18 biomolecule ion based on its trajectory.

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RESIDUES OF LYSOZYME FOLLOWING 10 MINUTE DIGEST  
ON AN Au/TRYPsin PROBE TIP

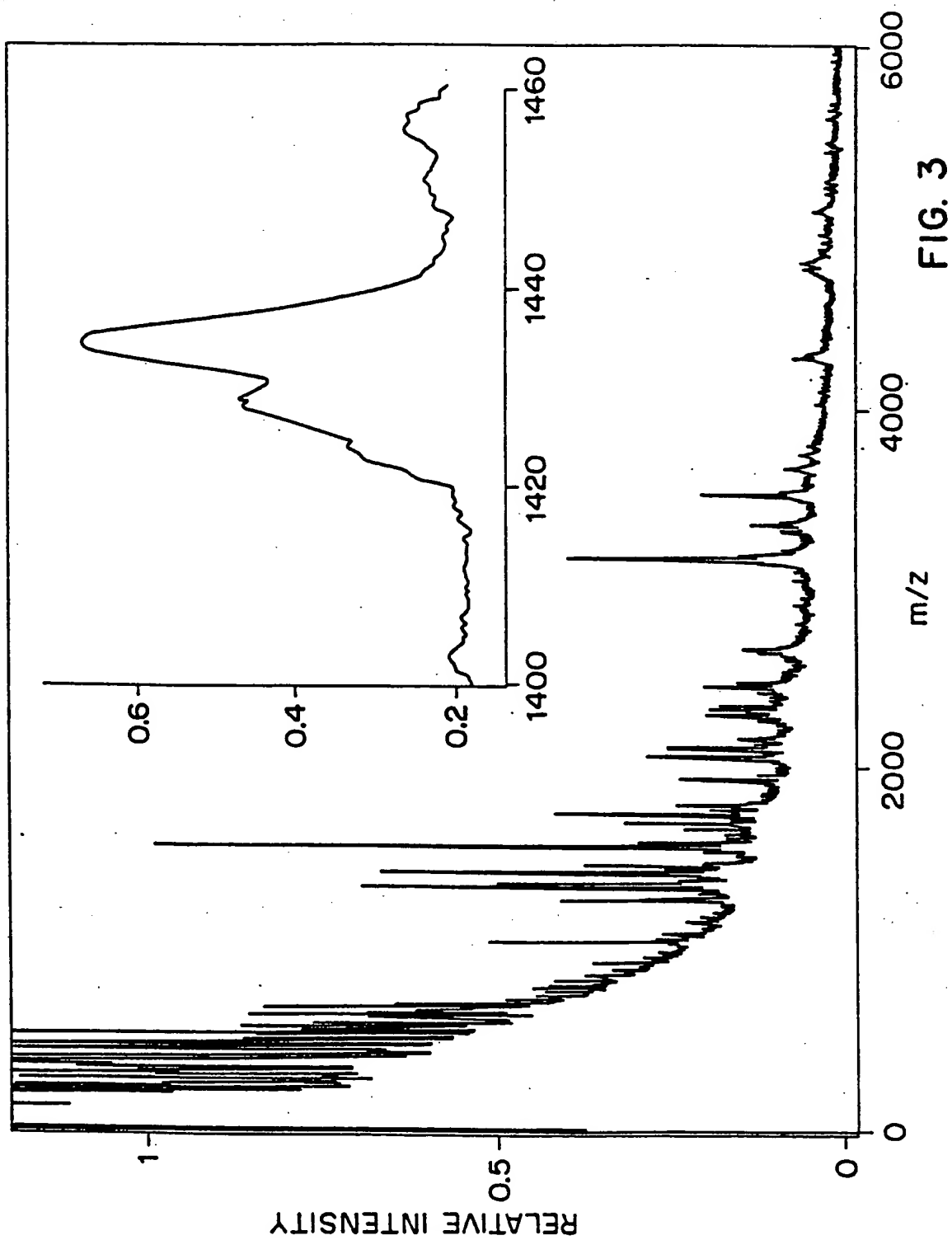
MASS <sub>(observed)</sub> Da	MASS <sub>(calculated)</sub> Da	$\Delta$ Da	RESIDUES
936.6	937.02	-0.42	62-68
1046	1045.54	0.46	117-125
1277.4	1276.64	0.76	22-33
1424.8	1424.78	0.02	1-13
1429.6	1428.64	0.96	34-45
1435.5	1434.63	0.87	62-73
1475.8	1474.75	1.05	117-129
1548	1546.78	1.22	113-125
1581	1580.96	0.04	1-14
1677.1	1676.90	0.20	98-112
1707.2	1707.04	0.16	115-129
1755	1754.86	0.14	46-61
1805.4	1805.08	0.32	97-112
1947.5	1947.19	0.31	98-114
1977.8	1977.33	0.47	113-129
2076.1	2075.37	0.73	97-114
2126	2125.38	0.62	15-33
2281.7	2281.57	0.13	14-33
2306.9	2304.69	2.21	97-116
2339.2	2336.02	2.48	74-96
2437.1	2436.88	0.22	1-21
2467.4	2464.90	2.50	74-97
2680.8	2679.94	0.86	22-45
3172.4	3171.41	0.99	46-73
3334.8	3333.84	0.96	97-125
3537.2	3535.68	1.52	15-45
3689.9	3692.05	-2.15	14-45
3754.4	3751.26	3.14	62-96
4125.5	4122.78	2.72	74-112
4418.4	4415.79	2.61	74-114
5098	5097.81	0.19	1-45
5273	5271.70	1.30	15-61
5488.3	5487.10	1.20	46-96

FIG. 2

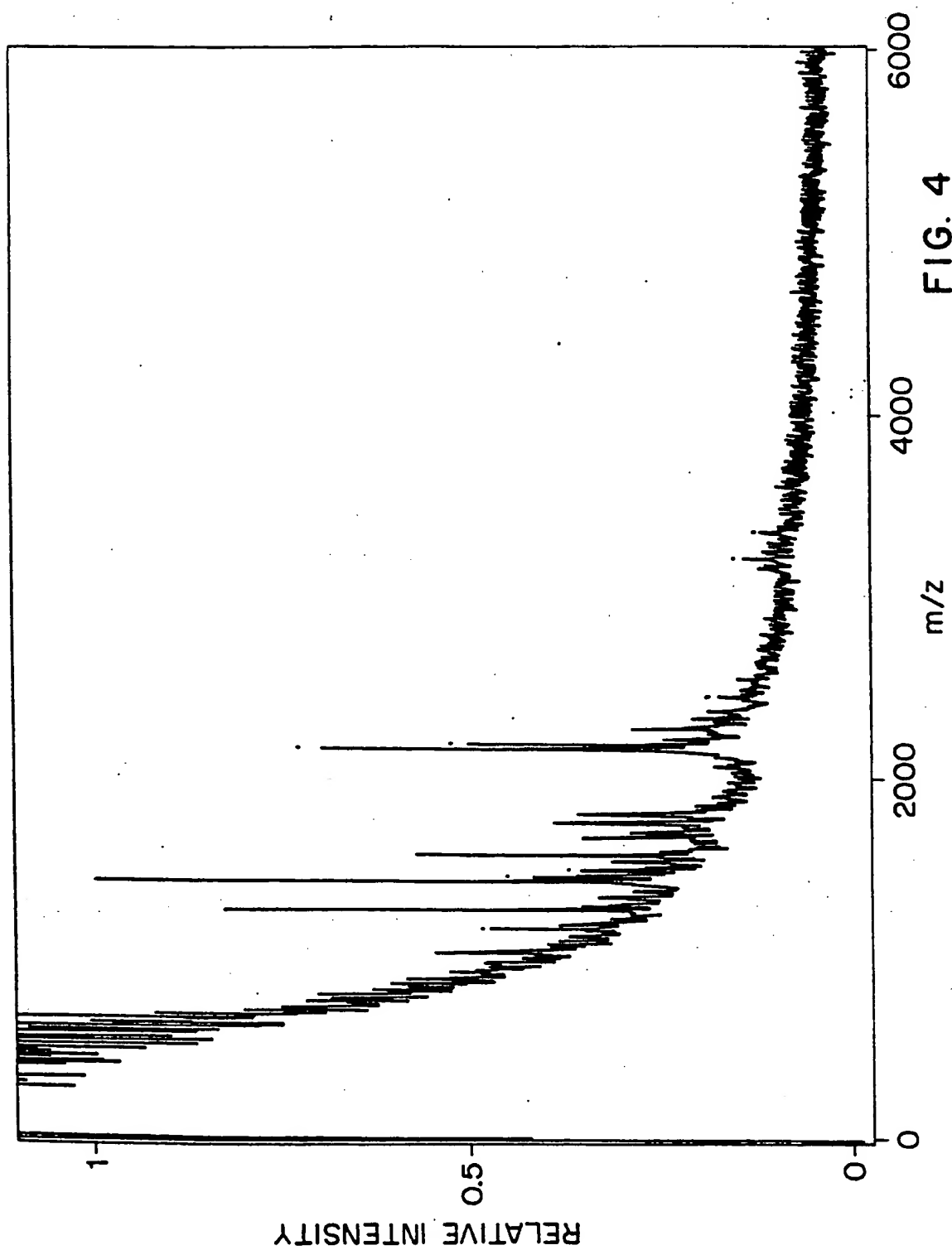
SUBSTITUTE SHEET (RULE 26)



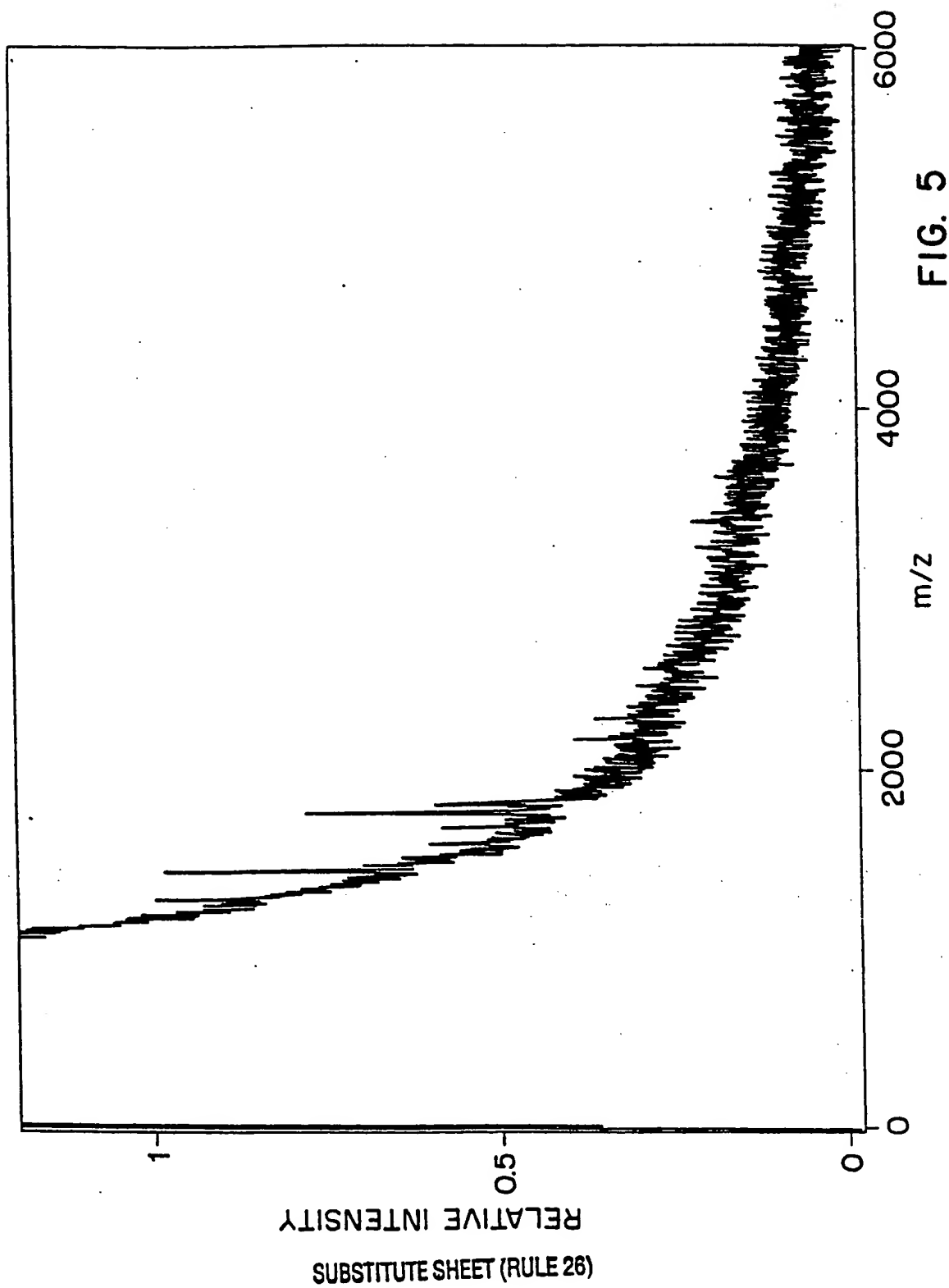
3/22



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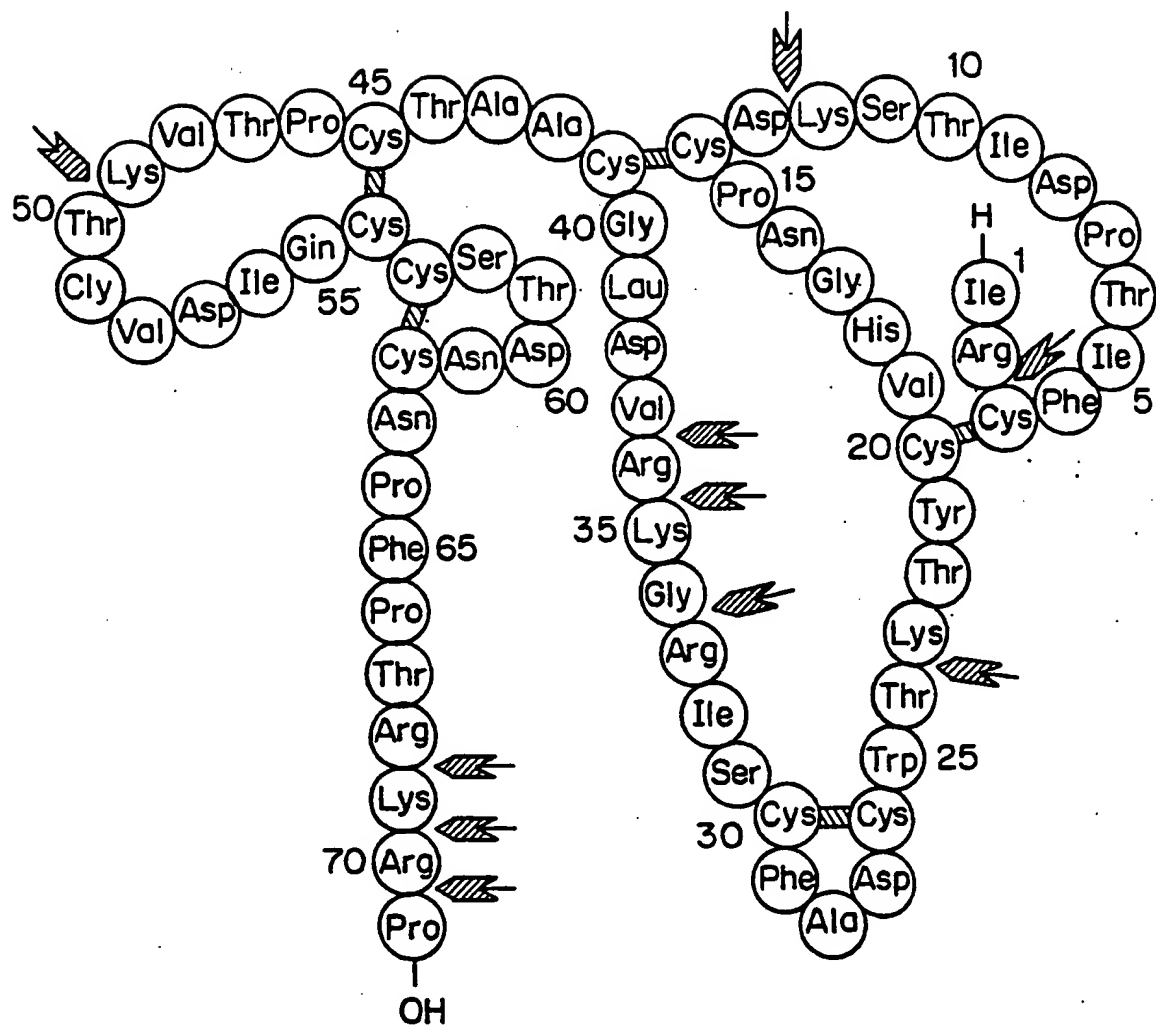
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RESIDUES OF LYSOZYME FOLLOWING 10 MINUTE DIGEST OF  
1 PMOL WITH FREE-TRYPSIN AND AGAROSE-IMMOBILIZED  
TRYPSIN

FREE TRYPSIN MASS(observed)Da	IMMOBILIZED TRYPSIN MASS(observed)Da	RESIDUES	AUTOLYSIS PRODUCT
935.3	936.5	62-68	
1046.1	1046.1	117-125	
1176.7			X
1277.4	1276.9	115-125	
1435.6	1434.9	62-73	
1458.9			X
1476.9		117-129	
1497.9			X
1548.8		13-125	
1582.4	1581.8	1-14	
1678.2	1677.6	98-112	
1709.4		115-129	
1756.7	1755.5	46-61	
1806.4	1805.8	97-112	
2166	2164.4		X
2190.6			X
2274	2274.8		
2471.5			X
3168		46-73	
3227.9			X
3372.5	3371.1		X

FIG. 6

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➡ = TRYPSIN CLEAVAGE SITE

FIG. 7

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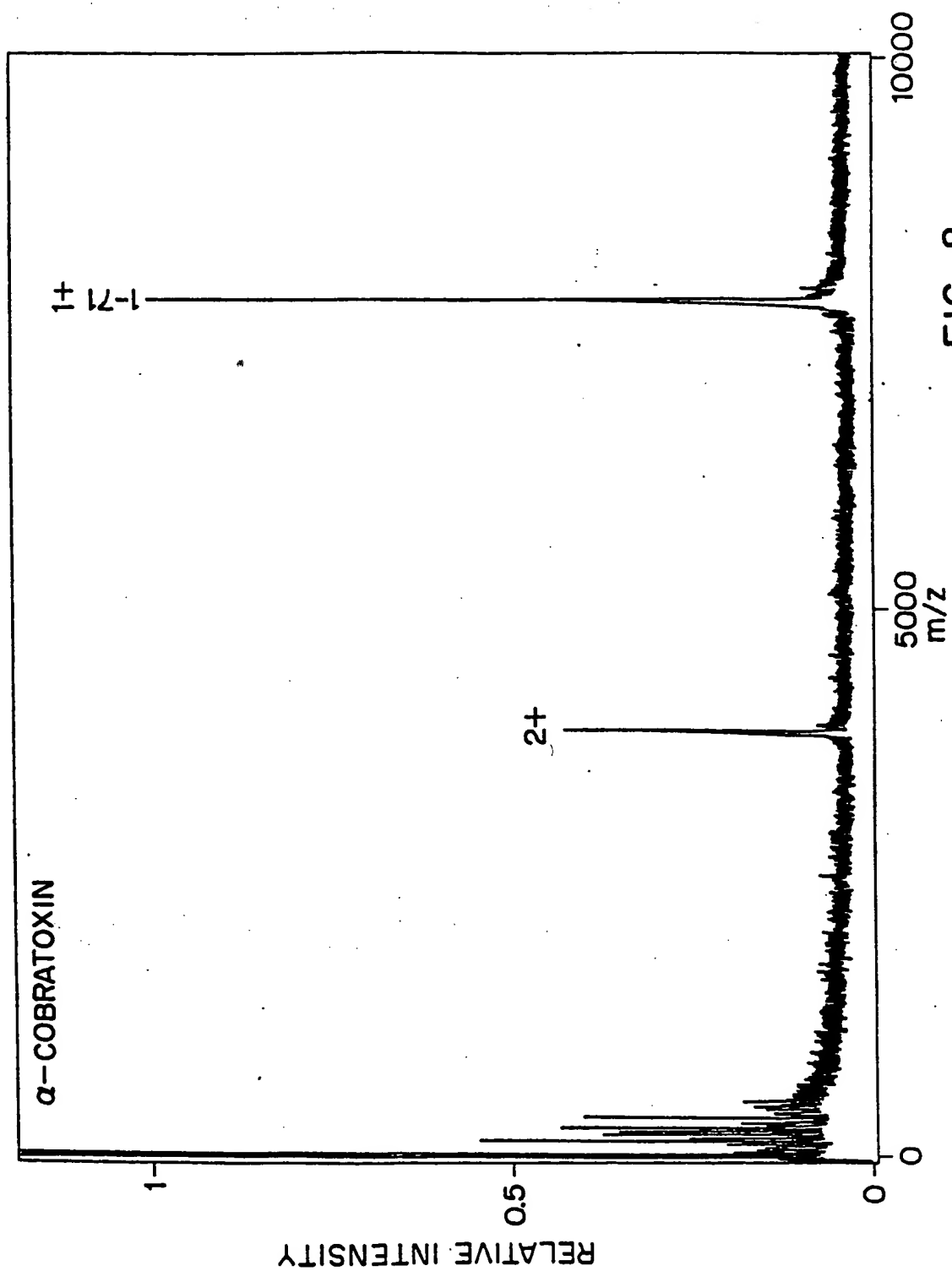


FIG. 8

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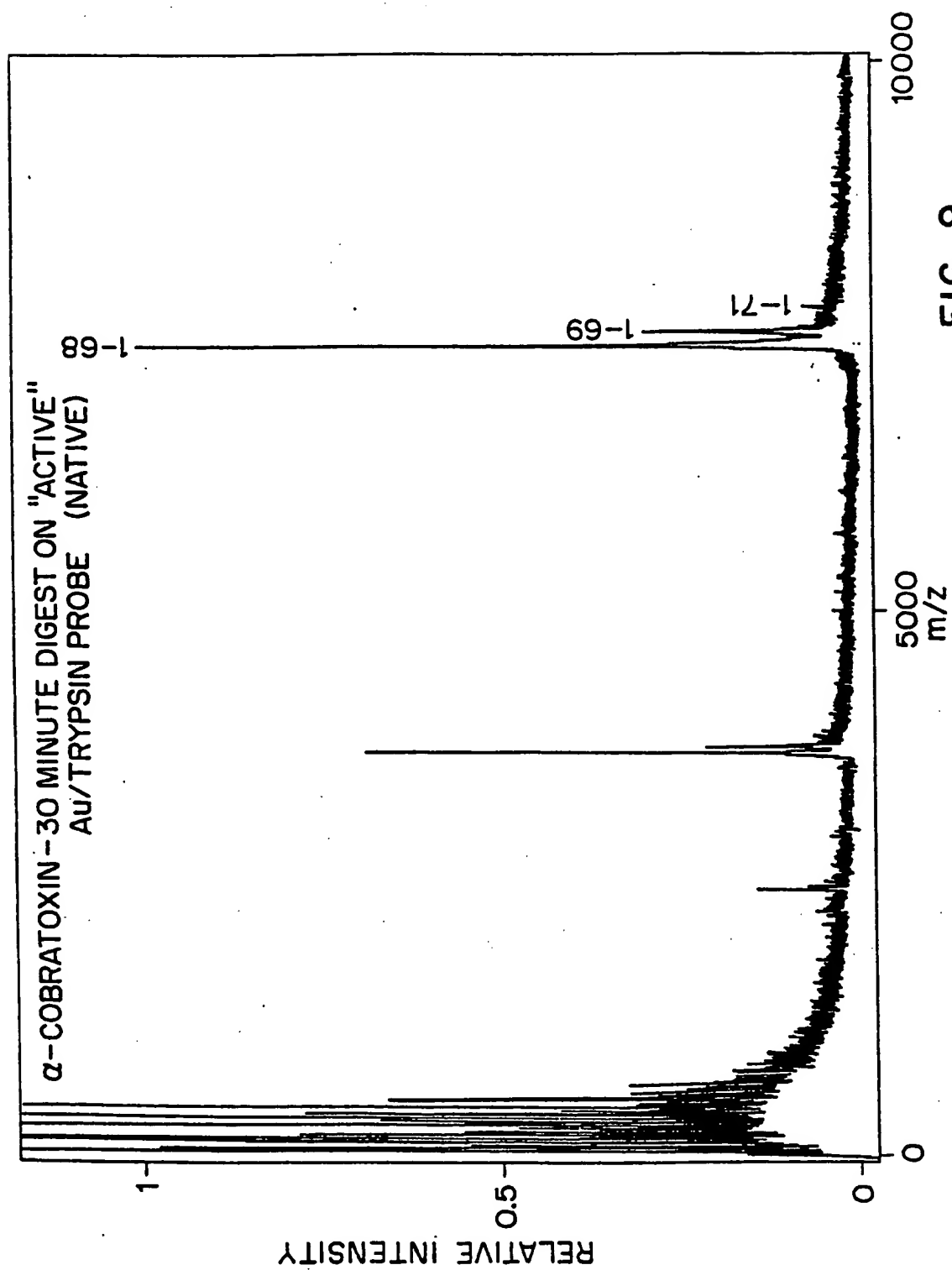


FIG. 9

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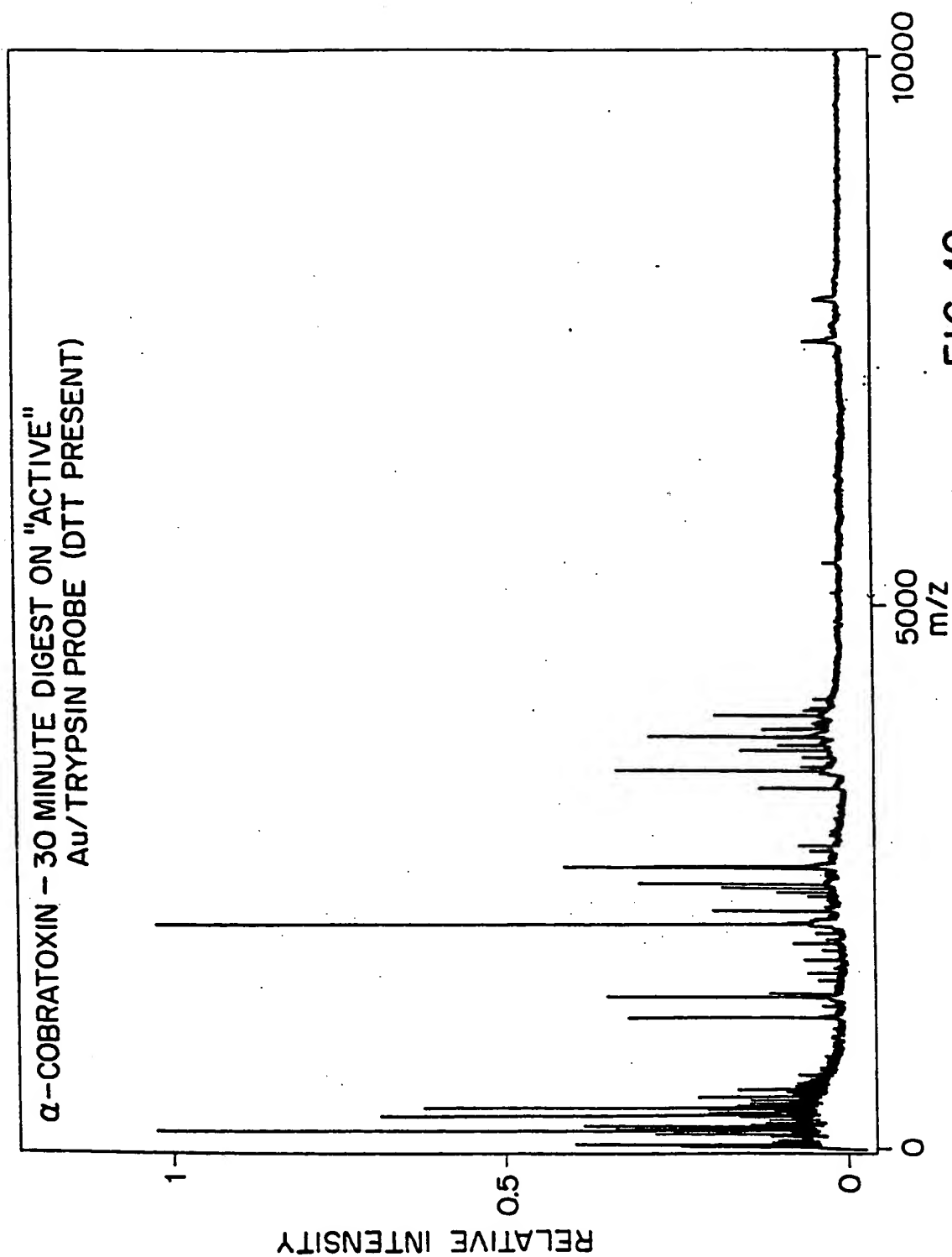


FIG. 10



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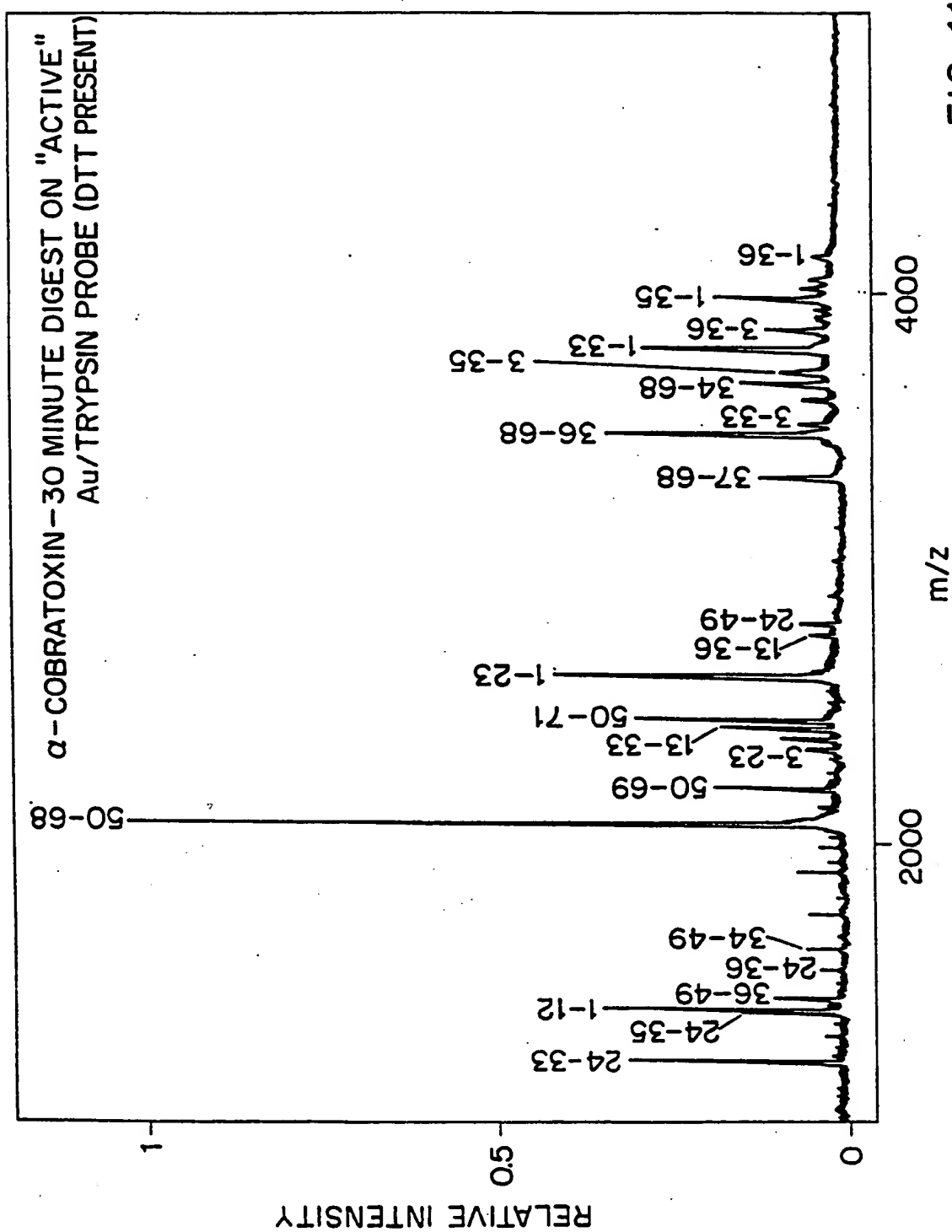


FIG. 11

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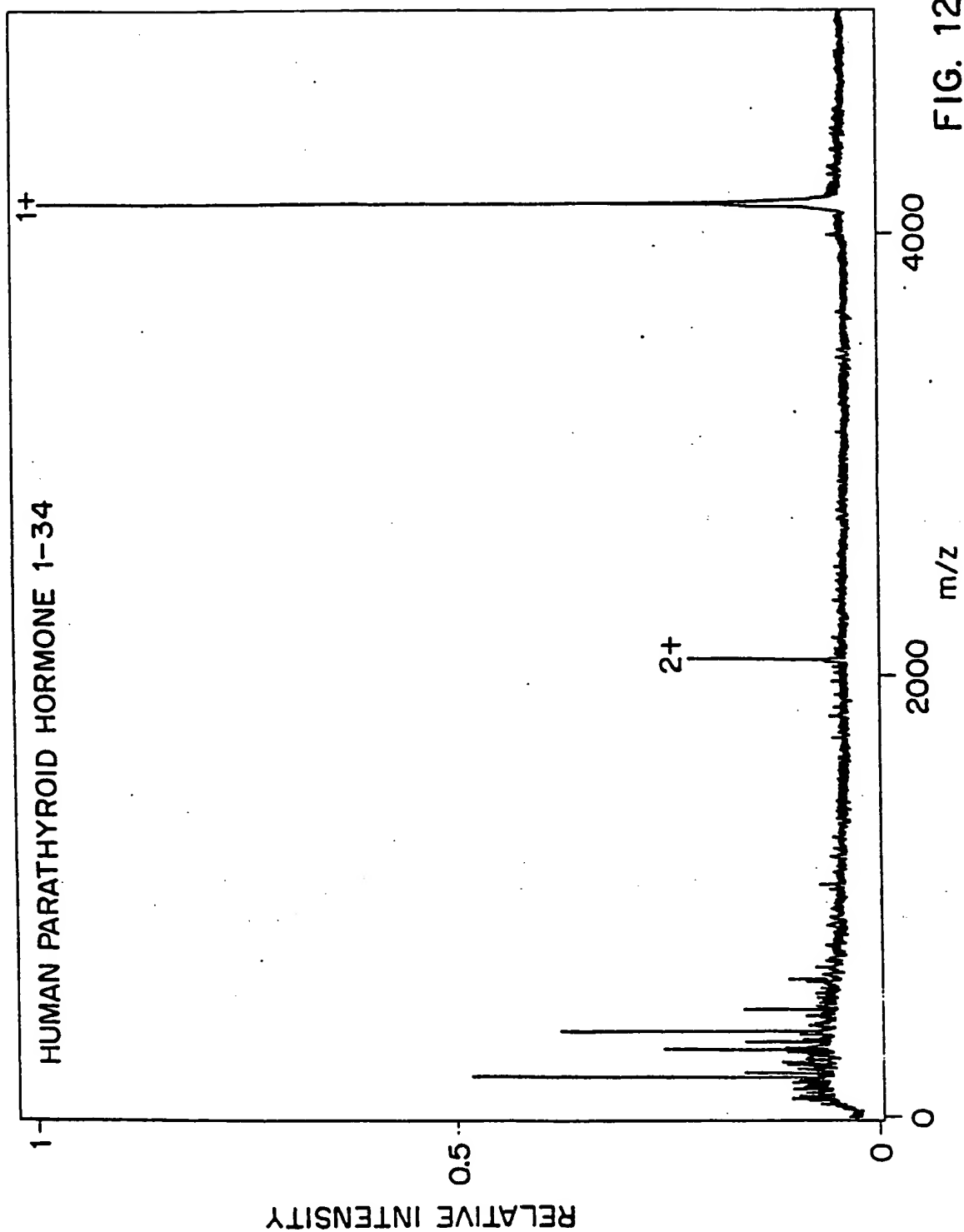
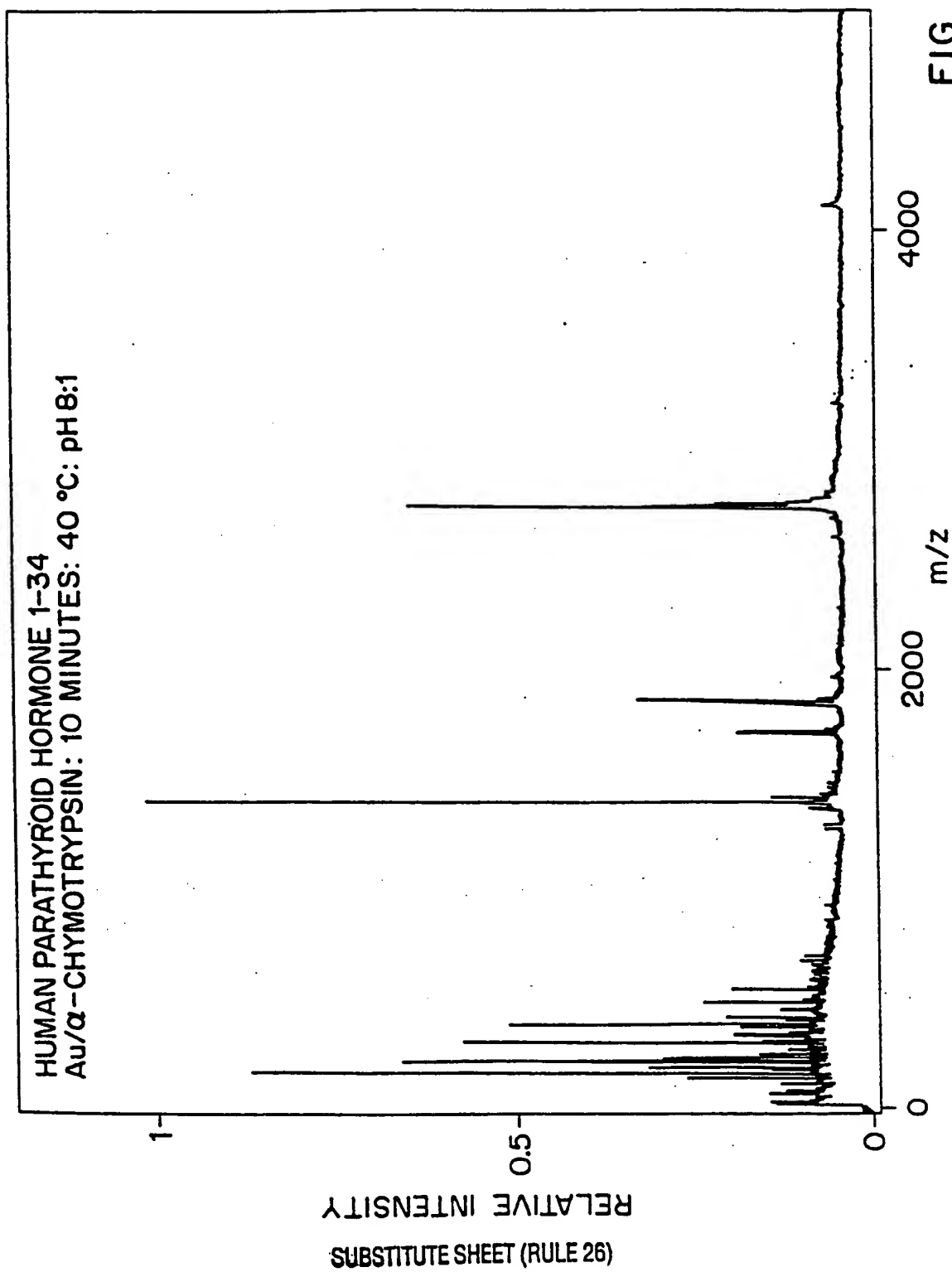


FIG. 12

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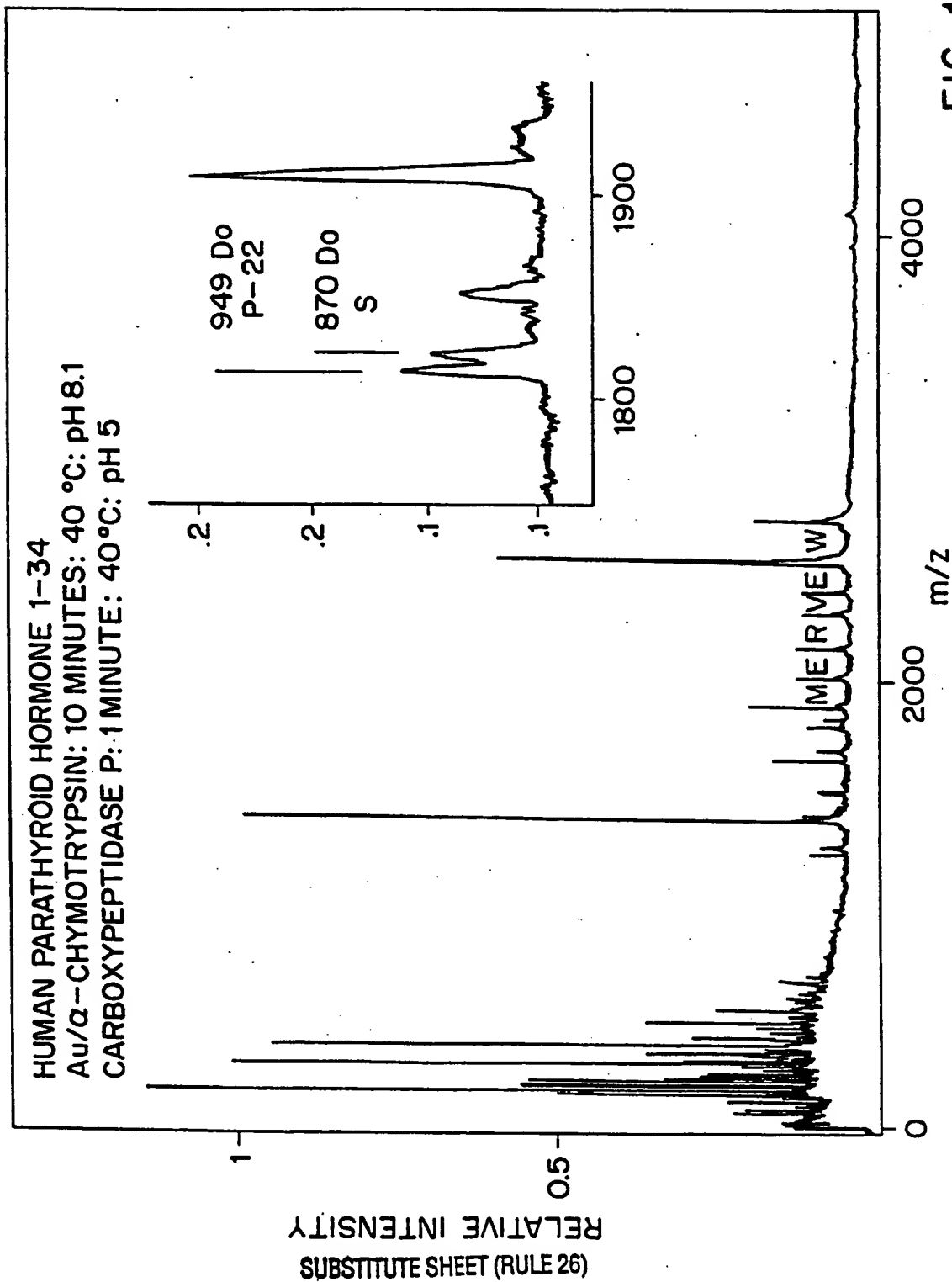


FIG. 14

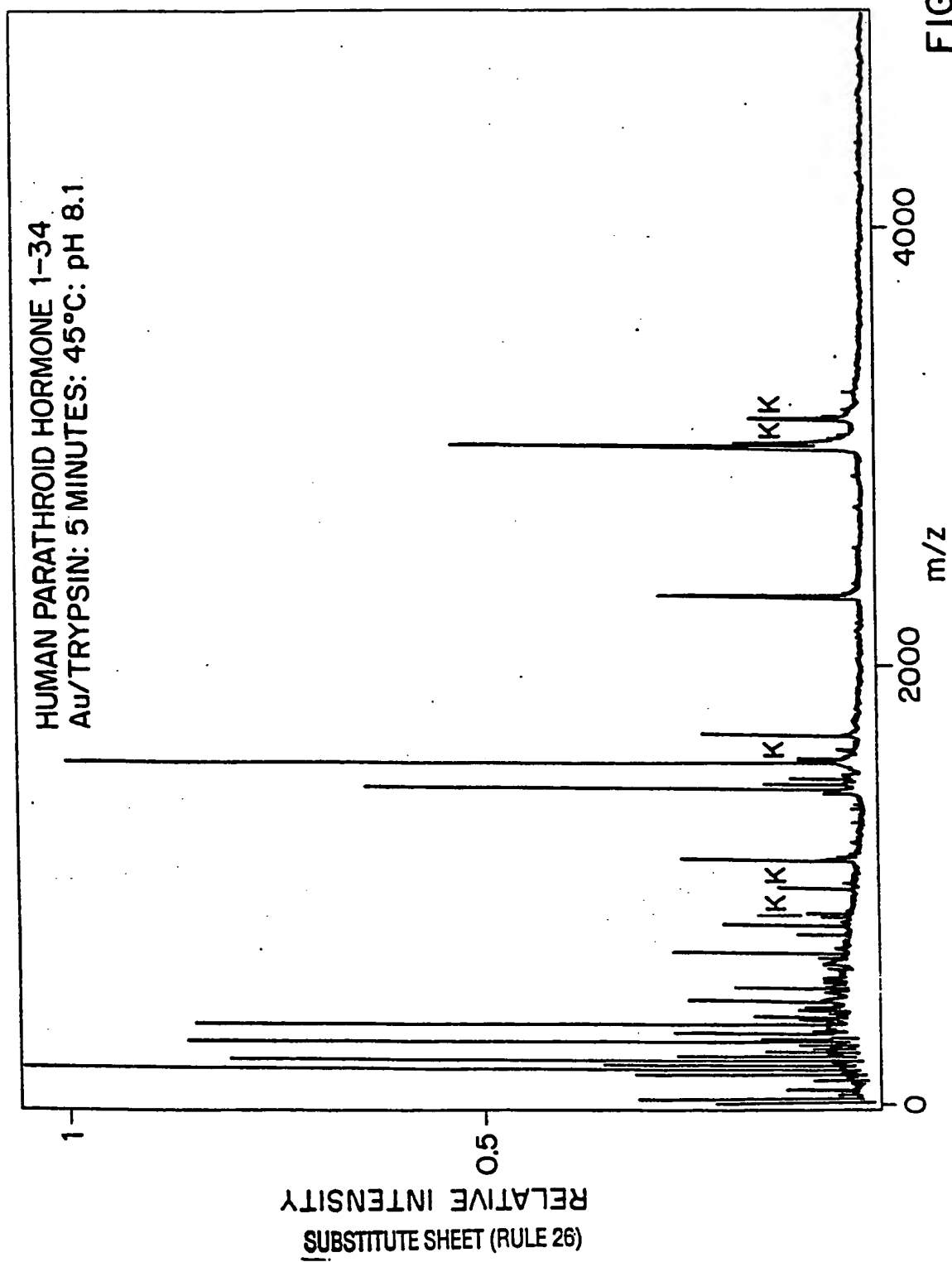
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hPTH Au/ $\alpha$ -CHYMOTRYPSIN/CARBOXYPEPTIDASE P

ENDOPEPTIDASE SIGNAL	LADDER SIGNAL	DIFFERENCE	RESIDUE	ERROR
2,738.7	2,738.9	0		
	2,552.4	186.5	W	+0.3
	2,423.5	128.9	E	-0.2
	2,324.3	99.2	V	+0.1
	2,168.1	156.2	R	0.0
	2,038.8	129.3	E	+0.2
	1,907.6	131.2	M	0.0
1,850.0				
	1,820.6	87.0	S	-0.1
1,706.2				
1,398.5				

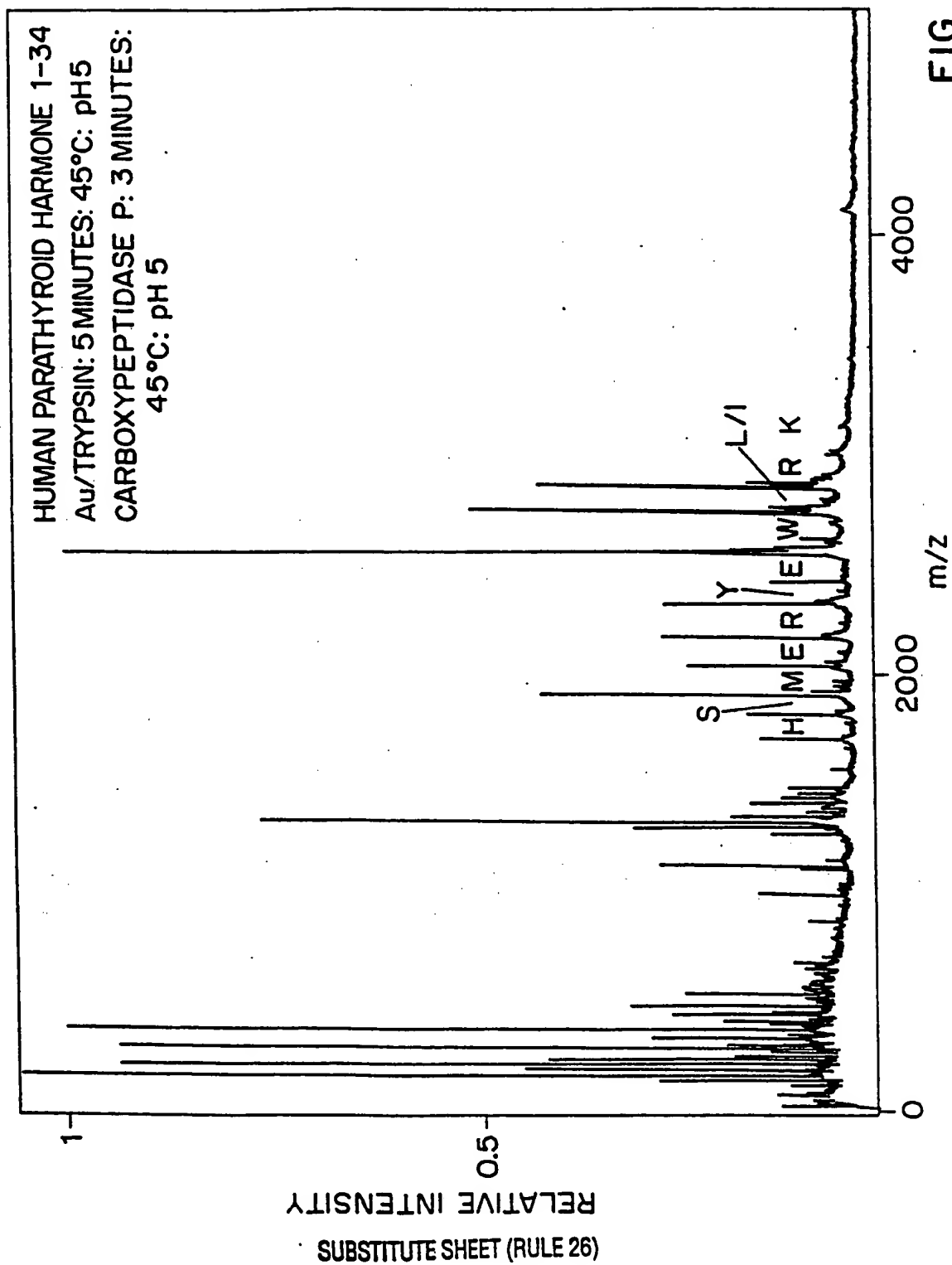
FIG. 15

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**FIG. 16**

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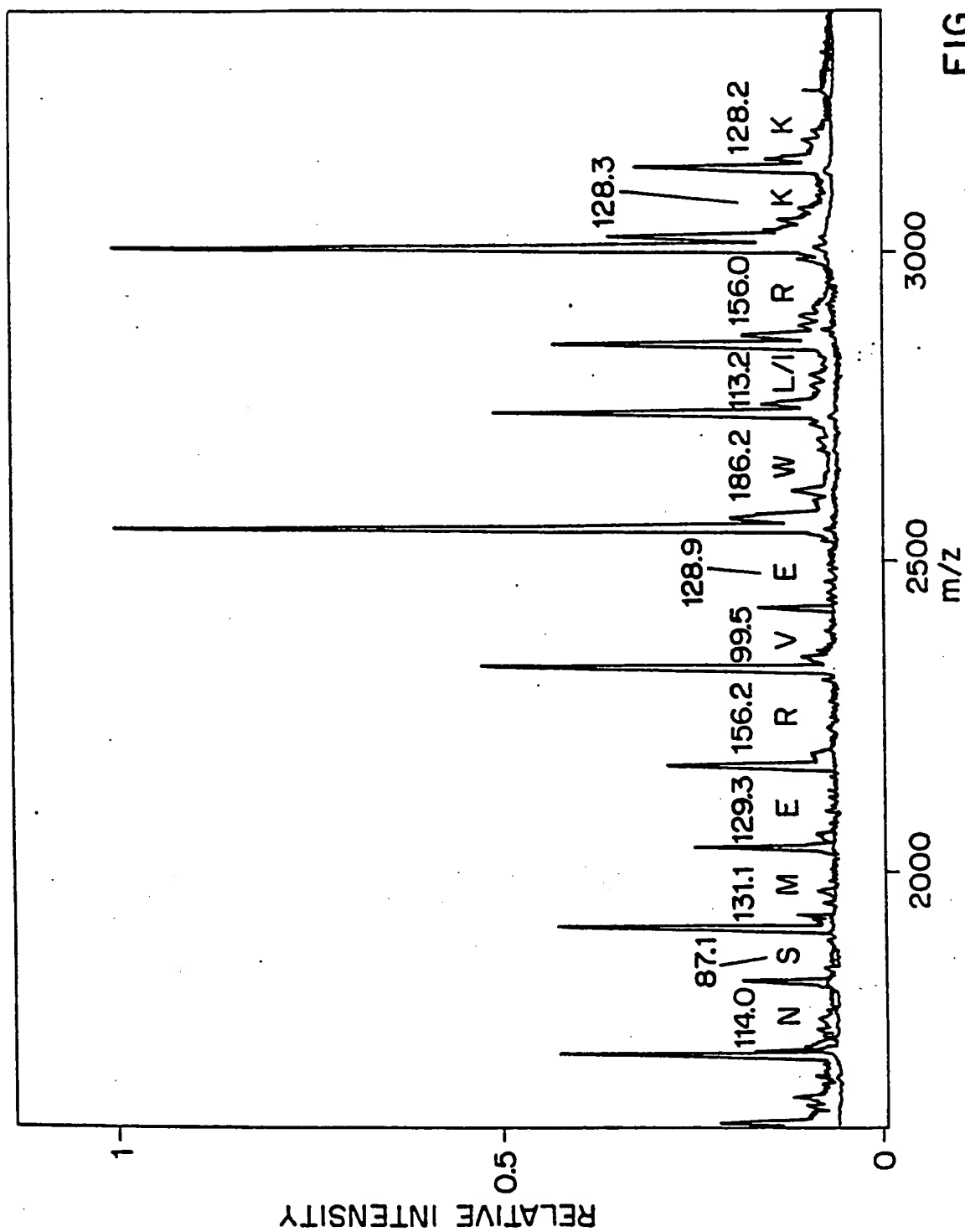
# hPTH Au/TRYPsin/CARBOXYPEPTIDASE P (\* DENOTES LYS-LYS)

ENDOPEPTIDASE SIGNAL	LADDER SIGNAL	DIFFERENCE	RESIDUE	ERROR
*3,264.3				
*3,136.1	3,136.1	128.2	K	0.0
*3,007.9	3,007.8	128.3	K	+0.1
	2,851.8	156.0	R	-0.2
	2,738.6	113.2	L/I	0.0
	2,552.4	186.2	W	0.0
	2,423.5	128.9	E	-0.2
2,323.8	2,324.0	99.5	V	+0.4
	2,167.8	156.2	R	0.0
	2,038.5	129.3	E	+0.2
	1,907.4	131.1	M	0.0
	1,820.3	87.1	S	0.0
	1,706.3	114.0	N	-0.1
1,698.2				
1,570.0		128.2	K	0.0
1,455.9				
*1,128.7				
*1,000.5		128.2	K	0.0
886.4				
*872.3		128.2	K	0.0
830.4				
702.3		128.1	K	-0.1

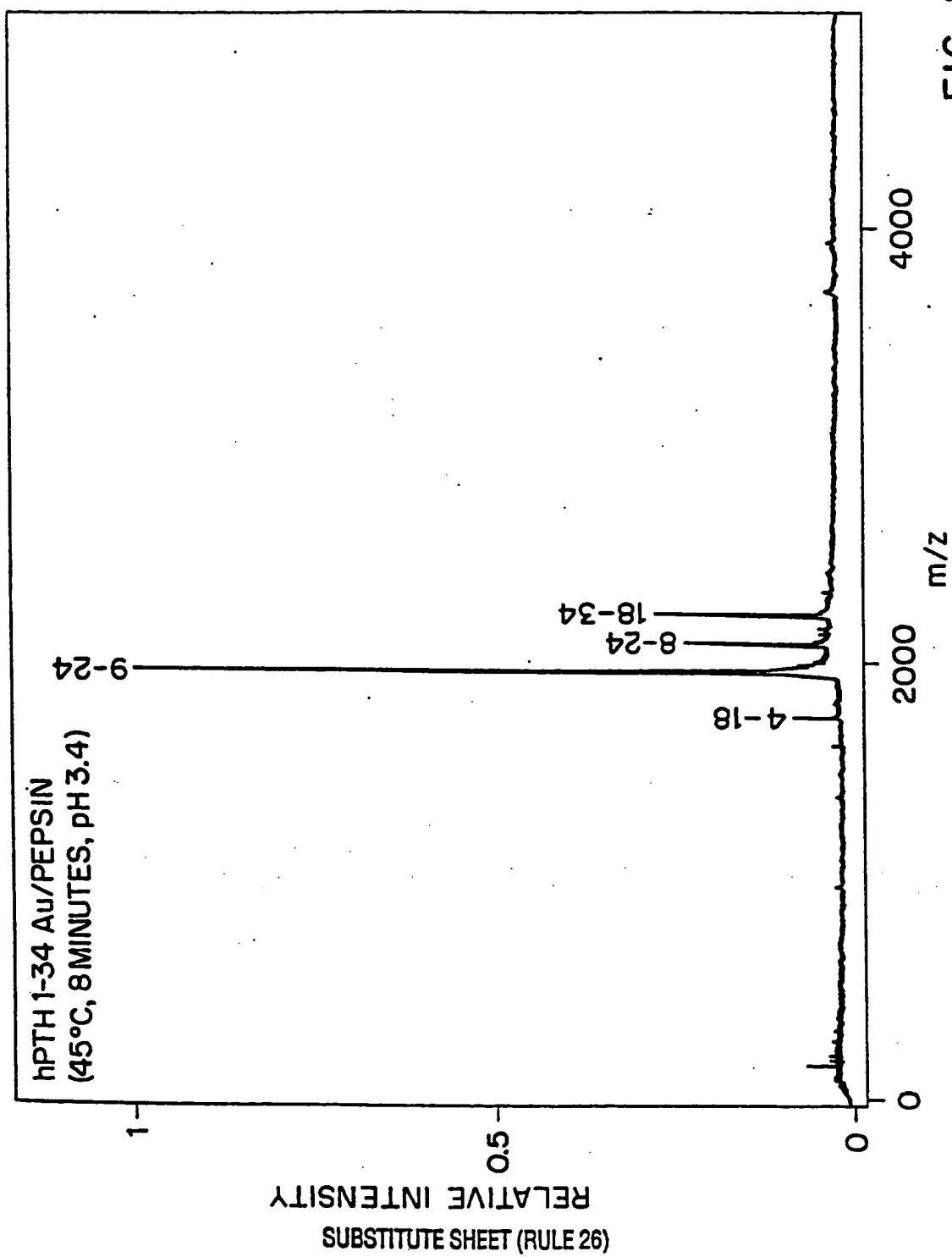
FIG. 18



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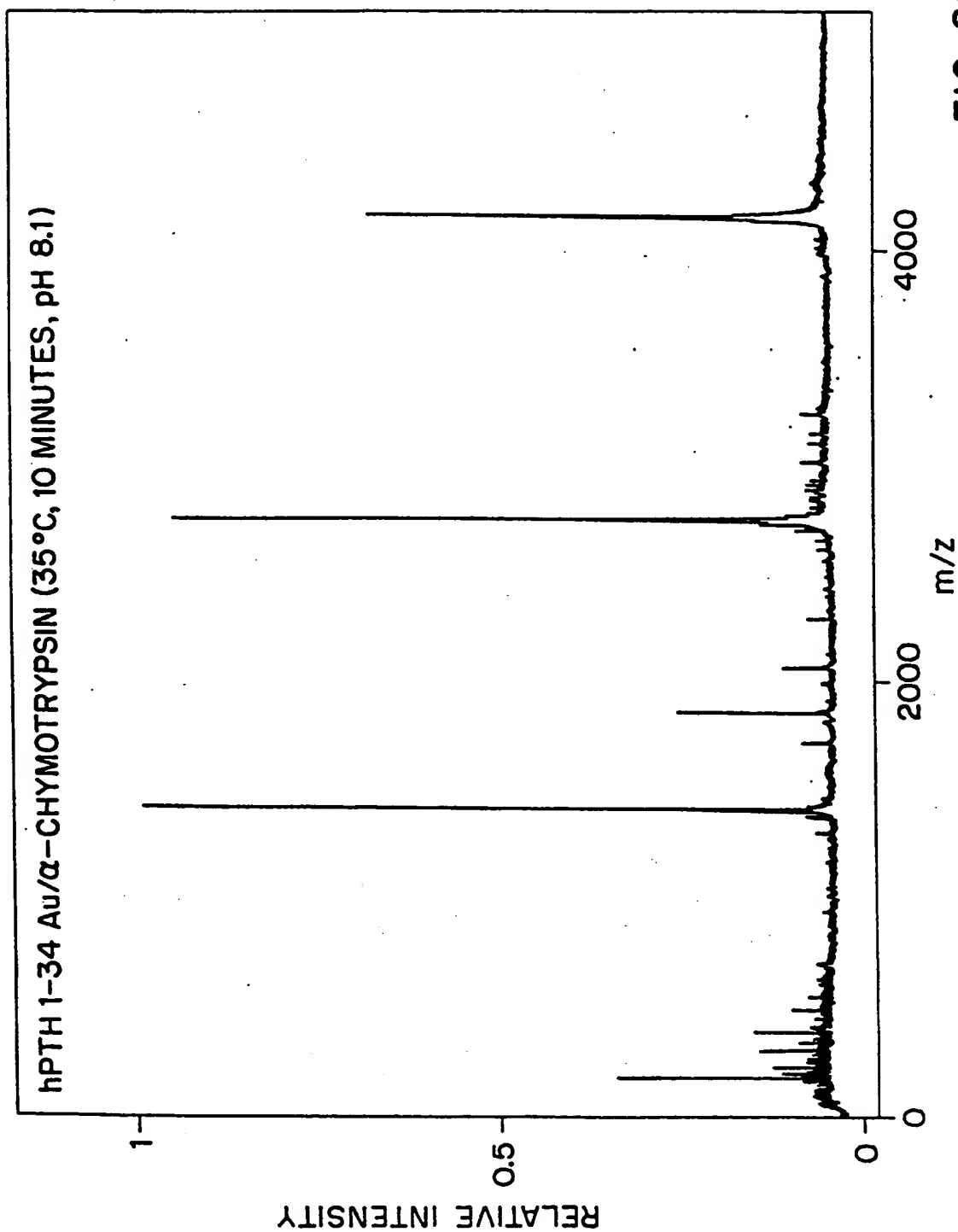


FIG. 21

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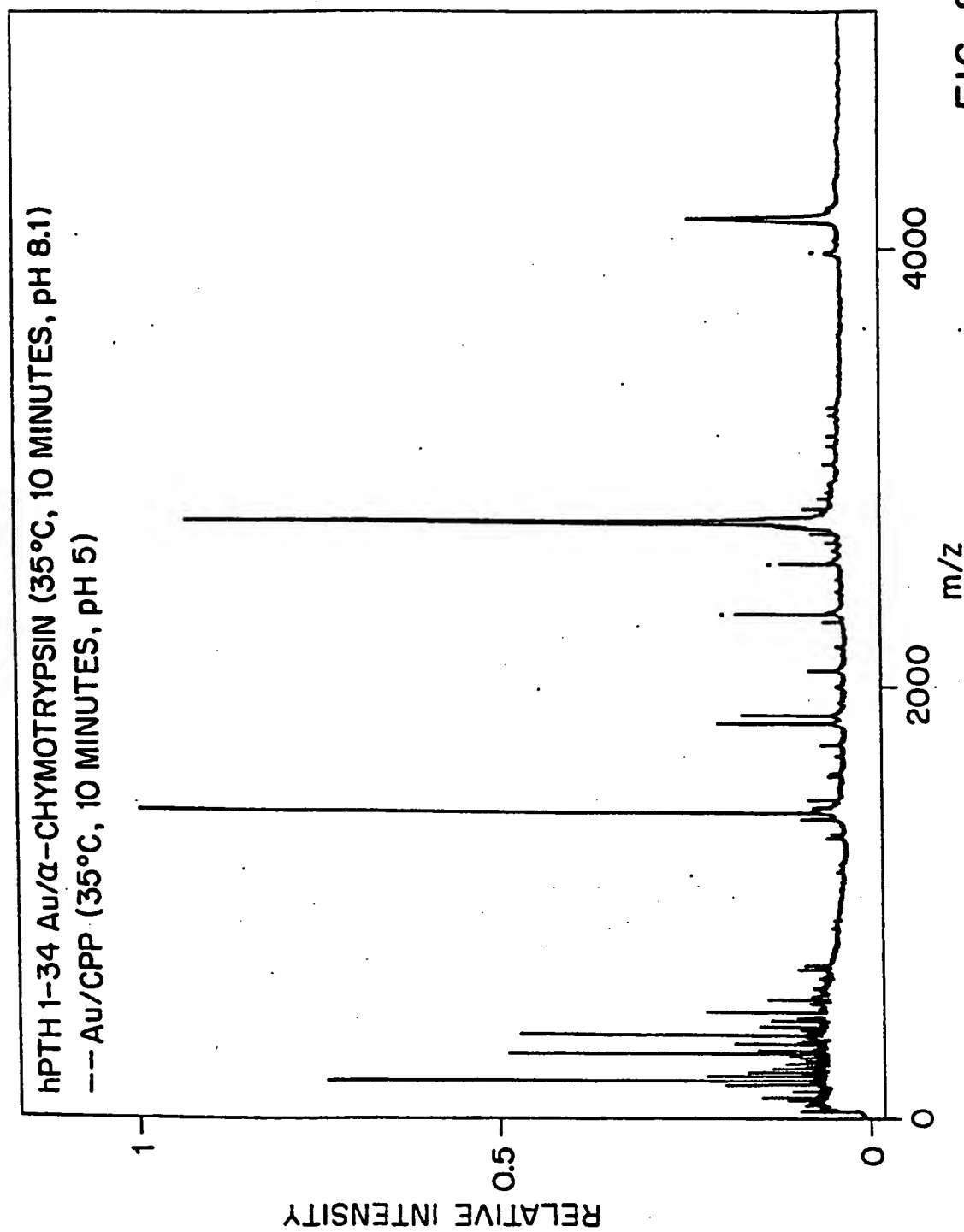


FIG. 22

## INTERNATIONAL SEARCH REPORT

 International application No.  
 PCT/US96/08994

## A. CLASSIFICATION OF SUBJECT MATTER

IPC(6) : C12N 11/00; C12Q 1/37, 1/68; H01J 17/26, 27/00, 49/00

US CL : 250/281, 288, 423P; 313/566; 435/6, 23, 24, 174

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

U.S. : 250/281, 288, 423P; 313/566; 435/6, 23, 24, 174

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)

APS, BIOSIS, CHEMISTRY ABSTRACTS, DERWENT

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of documents, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 94/28418 A1 (BAYLOR COLLEGE OF MEDICINE) 08 December 1994, see pages 4, 10, 17, 18, 26, 30-32, 48.	1-41
A	US 4,988,879 A (ZARE et al.) 29 January 1991, see entire document.	1-41
A, P	DOGRUEL, et al. Rapid Tryptic Mapping Using Enzymatically Active Mass Spectrometer Probe Tips. Analytical Chemistry. 01 December 1995, Vol. 67, No. 23, pages 4343-4348.	1-41

☐ Further documents are listed in the continuation of Box C.
 ☐ See patent family annex.

* Special categories of cited documents:	"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
"A" document defining the general state of the art which is not considered to be part of particular relevance	"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
"E" earlier document published on or after the international filing date	"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art
"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	"A" document member of the same patent family
"O" document referring to an oral disclosure, use, exhibition or other means	
"P" document published prior to the international filing date but later than the priority date claimed	

Date of the actual completion of the international search

19 JULY 1996

Date of mailing of the international search report

30 SEP 1996

 Name and mailing address of the ISA/US  
 Commissioner of Patents and Trademarks  
 Box PCT  
 Washington, D.C. 20231

Facsimile No. (703) 305-3230

Authorized officer

RALPH GITOMER

Telephone No. (703) 308-0196

# INTERNATIONAL SEARCH REPORT

International application No.  
PCT/US96/08994

## Box I Observations where certain claims were found unsearchable (Continuation of item 1 of first sheet)

This international report has not been established in respect of certain claims under Article 17(2)(a) for the following reasons:

1. ☐ Claims Nos.:  
because they relate to subject matter not required to be searched by this Authority, namely:
  
2. ☐ Claims Nos.:  
because they relate to parts of the international application that do not comply with the prescribed requirements to such an extent that no meaningful international search can be carried out, specifically:
  
3. ☐ Claims Nos.:  
because they are dependent claims and are not drafted in accordance with the second and third sentences of Rule 6.4(a).

## Box II Observations where unity of invention is lacking (Continuation of item 2 of first sheet)

This International Searching Authority found multiple inventions in this international application, as follows:

Please See Extra Sheet.

1. ☒ As all required additional search fees were timely paid by the applicant, this international search report covers all searchable claims.
2. ☐ As all searchable claims could be searched without effort justifying an additional fee, this Authority did not invite payment of any additional fee.
3. ☐ As only some of the required additional search fees were timely paid by the applicant, this international search report covers only those claims for which fees were paid, specifically claims Nos.:
  
4. ☐ No required additional search fees were timely paid by the applicant. Consequently, this international search report is restricted to the invention first mentioned in the claims; it is covered by claims Nos.:

Remark on Protest

- ☐ The additional search fees were accompanied by the applicant's protest.  
☐ No protest accompanied the payment of additional search fees.

## INTERNATIONAL SEARCH REPORT

International application No.

PCT/US96/08994

### BOX II. OBSERVATIONS WHERE UNITY OF INVENTION WAS LACKING

This ISA found multiple inventions as follows:

This application contains the following inventions or groups of inventions which are not so linked as to form a single inventive concept under PCT Rule 13.1. In order for all inventions to be examined, the appropriate additional examination fees must be paid.

Group I, claim(s) 1-9, drawn to a sample presentation apparatus and method of making it.

Group II, claim(s) 10-41, drawn to a mass spec and method of using it.

The inventions listed as Groups I and II do not relate to a single inventive concept under PCT Rule 13.1 because, under PCT Rule 13.2, they lack the same or corresponding special technical features for the following reasons: The apparatus and method of Group I is independent and does not require any feature of the apparatus and method of Group II. 37 CFR 1.475(d).